Vacuum Science and Technology for Accelerator Vacuum Systems

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SESSION 2: VACUUM INSTRUMENTATION

- Overview of total pressure gauges
- Direct pressure gauges
- Indirect pressure gauges
- Partial pressure gauges
- Gauge selection considerations
Vacuum Pressure Measurements

- **The Ideal Gas Law** - the foundation of vacuum measurements:

  \[ P = n k T \]

- **Direct pressure gauges** - Those gauges directly sense force per unit area. The direct gauges give 'true' measure of pressure, independent of gas types, and they may be used as primary pressure standards.

- **Indirect pressure gauges** - Those gauges explore the relations between certain physical properties (such as ionizations, viscosity, thermal energy) and the gas density. The indirect gauges are gas-type dependent, and require calibrations.
Total Pressure Gauges
Direct Gauges at a Glance

Direct Gauges
(Displacement of a wall)

Solid Wall
- Diaphragm
- Bourdon Type
- Capacitance Diaphragm

Liquid Wall
- McLeod
- U-Tube Manometer
Indirect Gauges at a Glance

Indirect Gauges
(Measurement of a gas property)

Charge Generation (Ionization)

Energy Transfer (Heat Loss)

Momentum Transfer (Viscosity)

Thermocouple  Pirani  CONVECTION

Cold Cathode  Hot Cathode

Penning  Inverted Magnetron  Triode  Bayard-Alpert  Extractor  Helmer, Bend Beam

Spinning Rotor
Vacuum Pressure Ranges

- In today’s scientific research and industrial processes, vacuum measurements cover over 17 decades of range, from atmospheric pressure \(10^5\) Pa down to \(10^{-12}\) Pa.
- For most applications, a combination of multiple types of gauges is needed.

<table>
<thead>
<tr>
<th>LOG P [Pa]</th>
<th>Hot Cathode Ionization Gauges</th>
<th>Cold Cathode Ionization Gauges</th>
</tr>
</thead>
<tbody>
<tr>
<td>-13</td>
<td></td>
<td></td>
</tr>
<tr>
<td>-11</td>
<td></td>
<td></td>
</tr>
<tr>
<td>-9</td>
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<td>-7</td>
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<td>-5</td>
<td></td>
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<tr>
<td>-3</td>
<td></td>
<td></td>
</tr>
<tr>
<td>-1</td>
<td></td>
<td></td>
</tr>
<tr>
<td>1</td>
<td></td>
<td></td>
</tr>
<tr>
<td>3</td>
<td></td>
<td></td>
</tr>
<tr>
<td>5</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

Gauges:
- Liq. Manometer
- McLeod
- Therm. Cond. GA
- Cap. Diaphragm GA
- Spinning Rotor Gauge
Pressure Units and Conversions

- Mercury manometers have been used since the earliest days of vacuum technology. Thus the mmHg, or Torr is commonly used pressure unit, especially in the US.
- The SI unit for pressure is Pascal = 1 N/m².
- mbar is also commonly used, mostly used in Europe (and 'allowed' in SI). 1 mbar = 1.00x10² Pa = 0.750 Torr

<table>
<thead>
<tr>
<th></th>
<th>Pa</th>
<th>mbar</th>
<th>Torr</th>
<th>In. Hg</th>
<th>PSI</th>
<th>atm</th>
</tr>
</thead>
<tbody>
<tr>
<td>Pa</td>
<td>1</td>
<td>1.00x10²</td>
<td>1.33x10²</td>
<td>3.39x10³</td>
<td>6.89x10³</td>
<td>1.01x10⁵</td>
</tr>
<tr>
<td>mbar</td>
<td>1.00x10⁻²</td>
<td>1</td>
<td>1.33</td>
<td>3.39x10¹</td>
<td>6.89x10¹</td>
<td>1.01x10³</td>
</tr>
<tr>
<td>Torr</td>
<td>7.50x10⁻³</td>
<td>7.50x10⁻¹</td>
<td>1</td>
<td>2.54x10¹</td>
<td>5.17x10¹</td>
<td>7.60x10²</td>
</tr>
<tr>
<td>In. Hg</td>
<td>2.95x10⁻⁴</td>
<td>2.95x10⁻²</td>
<td>3.94x10⁻²</td>
<td>1</td>
<td>2.04</td>
<td>2.99x10¹</td>
</tr>
<tr>
<td>PSI</td>
<td>1.45x10⁻⁴</td>
<td>1.45x10⁻²</td>
<td>1.93x10⁻²</td>
<td>4.91x10⁻¹</td>
<td>1</td>
<td>1.47x10¹</td>
</tr>
<tr>
<td>atm.</td>
<td>9.87x10⁻⁶</td>
<td>9.87x10⁻⁴</td>
<td>1.32x10⁻³</td>
<td>3.34x10⁻²</td>
<td>6.80x10⁻²</td>
<td>1</td>
</tr>
</tbody>
</table>
Liquid Manometers – U-Tubes

- Simplest direct gauge. Mercury or oil are often used.

\[ P_1 - P_2 = h \rho g \]

- Main source of error is in measuring \( h \)

- Elaborated methods developed to measure \( h \), using optical or ultra-sonic interferometer, to achieve accuracy of 1.4 mPa in range of 1 Pa to 100 kPa in NIST, as US primary pressure standard
NIST Ultrasonic Interferometer

*) Jay H. Hendricks *, Douglas A. Olson,
Measurement, 43 (2010) 664–674
Liquid Manometers – McLeod Gauge

- McLeod gauge extends U-Tube manometer range using Boyle’s law. It is the primary pressure standard in the range of $10^{-2} \sim 10^3$ Pa.
Mechanic Diaphragm Gauges

- Direct gauge, independent of gas types
- Leybold DIAVAC 1000
- Range: 1 ~ 1000 mbar
Capacitance Diaphragm Gauges

- Commercial CDG systems can measure pressure ranges from $0.1 \sim 10^5$ Pa, independent of gas types.
- Usually a sensor can only cover 3~4 decades of pressure, with accuracy $\pm 0.5\%$. CDG system with temperature control can provide accuracy and stability $\pm 0.05\%$.
- CDGs are commonly used in thin film depositions.
- Main sources of errors are electronic drifts and diaphragm hysteresis.

$$C = \varepsilon_0 KA/s$$

Differential pressure cause change in spacing, $S$. 

![Diagram](image.png)
Capacitance Diaphragm Gauge – Example

INFICON’s CDGs

- All-ceramic diaphragm
- Wide ranges of pressure available
- Temperature compensated (stabilized)

Measurement ranges

<table>
<thead>
<tr>
<th>Full Scale (Torr)</th>
<th>0.2/0.4% Accuracy</th>
</tr>
</thead>
<tbody>
<tr>
<td>0.1</td>
<td></td>
</tr>
<tr>
<td>1</td>
<td></td>
</tr>
<tr>
<td>10</td>
<td></td>
</tr>
<tr>
<td>100</td>
<td></td>
</tr>
<tr>
<td>1000</td>
<td></td>
</tr>
</tbody>
</table>

0.00001 0.0001 0.001 0.01 0.1 1 10 100 1000
Pressure (Torr)
Indirect Vacuum Gauge Overview

- Indirect gauges measure pressure by relating certain physical properties to the gas density.
- Three major types indirect gauges are commonly used.
  - Thermal conductivity
  - Viscosity
  - Ionization
- Indirect gauges are gas type dependent
- Most commercially available indirect gauges are calibrated to nitrogen, thus relative calibrations of other gases to nitrogen is needed (and often supplied by the gauge manufacturers.)
Thermal Conductivity Gauges (1)

- A hot wire in a gaseous environment loses heat (thermal energy) in three ways: (1) radiation, $W_R$, (2) conduction to supports, $W_C$, and (3) transfer by the gas molecules, $W_G$.

\[
W_T = W_R + W_C + W_G
\]

- $W_G$ is pressure dependent, the base for these gauges.
- The heat gas transfer is approximately proportional to $m^{-1/2}$, and molecular compositions (atomic, diatomic, triatomic, etc.), thus is gas type dependent.
- $W_R$ and $W_C$ are independent of gas pressure, which determine the useful range of the gauges.
Heat transfer may be divided into three regimes, based the pressure (mean-free length, $\lambda$) 
(i) $\lambda \gg d_{\text{wire}}$; (ii) intermediate; (iii) $\lambda \ll d_{\text{wire}}$

(i) Heat transfer insignificant as useful for pressure measurement 
(iii) Gas heated by the hot wire may return energy back to wire
Thermocouple Gauge

- Constant current through the heater (sensor).
- TC junction measures temperature changes (due to gas heat transfer).
- Working range: $10^{-2} \sim 1$ torr
- Slow response time
Pirani Gauge

- In Pirani gauge, the heated filament constitutes an arm of a Wheatstone bridge. The Wheatstone bridge is balanced at high vacuum. Any gas heat transfer at higher pressure induces imbalance of the bridge.
- Working range: $10^{-4}$ ~ 100 torr.
- There are two common modes of operations.

**Constant Temperature Mode**

Adjusting heating current to maintain constant temperature (thus the resistance) to keep bridge balanced. The heating current is related to the pressure.

**Constant Voltage Mode**

Measure pressure with the changes in the imbalance current.
Convection Enhanced Pirani Gauge – Convectron

- Similar principle to Pirani gauge
  - Conductive heat loss (10^{-3} Torr to ~100 Torr)
  - Adds convective heat loss (~100 Torr to 1000 Torr)
- Improved temperature compensation.
- Gold plated tungsten sensor.
- Sensitive to mounting orientation
## Commercial Convection Gauges

**275 Convection Gauge**

### Technical Data

<table>
<thead>
<tr>
<th><strong>Range</strong></th>
<th>From atmosphere to $10^{-4}$ Torr, $(10^{-2}$ Pascal)</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Sensor Material</strong></td>
<td>Gold-plated tungsten</td>
</tr>
<tr>
<td><strong>Other materials exposed to gas</strong></td>
<td>304 stainless steel, borosilicate glass, Kovar, alumina, NiFe alloy, polyimide</td>
</tr>
<tr>
<td><strong>Internal Volume</strong></td>
<td>40 cm$^3$ (2.5 in$^3$)</td>
</tr>
<tr>
<td><strong>Operating Temperature</strong></td>
<td>0 °C to 50 °C ambient, non-condensing</td>
</tr>
<tr>
<td><strong>Bakeout Temperature</strong></td>
<td>150 °C maximum, non-operating, cable disconnected</td>
</tr>
<tr>
<td><strong>Connection</strong></td>
<td>1/8 inch NPT 1/2 inch tubulation</td>
</tr>
<tr>
<td><strong>Weight</strong></td>
<td>85 grams (3 ounces)</td>
</tr>
</tbody>
</table>

### 317 Convection-Enhanced Pirani Pressure Vacuum Sensors

(1.0x10$^{-3}$ to 1000 Torr)
Bakeable to 250°C
**Convectron Gauge Features**

**Good**
- ✓ Wide measurement range
- ✓ Long-term reliability (some sensors operational over 30 years in CESR)
- ✓ Low cost, low maintenance
- ✓ Relative fast response

**Not So Good**
- ❗ Gas dependent, often in complicated manners
- ❗ Not suitable for corrosive applications
- ❗ Orientation dependent (>10 Torr)
Spinning Rotor Gauge – Principle

- A spinning spherical rotor, suspended, in a gas at low pressure is slowed by interacting with the gas, through momentum transfer (or molecular drag).
- In the operational range (10^{-5} to 10^{-2} Pa), the deceleration by the molecular drag is proportional to the molecule density.
- SRG is gas type dependent.

\[ P = \frac{1}{5} \frac{a \rho}{\sigma} \frac{\sqrt{2\pi k T}}{\sqrt{m}} \left( -\frac{d\omega}{dt} - \frac{2\alpha}{\omega} \frac{dT}{dt} - RD \right) \]

\( a \) - rotor diameter; \( \rho \) - rotor density;
\( \sigma \) - gas accommodation coefficient;
\( m \) - gas molecule mass; \( \alpha \) - rotor C.T.E.
\( RD \) - residual drag (eddy current)
Spinning Rotor Gauge Structure

- **R** – rotor (440 stn. stl.)
- **V** – vacuum enclosure
- **M** – permanent magnets
- **A** – 2x pickup & axial control coils
- **L** – 4x lateral damping coils
- **D** – 4x drive coils
- **P** – 2x rotation sensing coils

\[ \omega = 410 - 400 \text{ rps} \]

- Low pressure limit: residual drags
- High pressure limit: non-linearity due to non-isotropic collisions with molecules
SRG maintains long term reproducibility, better than 1% over 7-year has been demonstrated. Thus SPGs are widely used as a transferable secondary pressure standards for gauge calibrations.

SRG does not ‘alter’ vacuum environment that measuring, as compared to ionization gauges.

SRG is sensitive to shock and vibration, as well as to changes in ambient temperature.

SRG is relatively slow in response time.
SRG Application – An Example

- Commercial SRG with easy to use electronics is available.
- We performed outgassing treatment to stainless steel to achieve ultra-low outgassing rate ($<10^{-14}$ torr-liter/s/cm$^2$).
- This ultra-low outgassing rate can only be measured by RoR method with SRG.

Test Pipes:
- $V = 29$-liter
- $A = 7,600$ cm$^2$

Ionization Gauges – General

➢ At pressures below $10^{-5}$ Torr (high vacuum), direct measurement of pressure is very difficult.
➢ Thermal conductivity gauges have exceeded their operational limits.
➢ Primary method for pressure measurement from $10^{-4}$ to $10^{-12}$ Torr is gas ionization & ion collection/measurement.
➢ These gauges can be generally divided into hot & cold cathode types.
➢ Most common high vacuum gauge today is the Bayard-Alpert and Inverted Magnetron.
Ionization Gauges – Principle

- Gas atoms and molecules are normally without charge or "neutral", they have equal numbers of protons and electrons.
- The neutrals may be 'ionized' via electron impact, to form ions.
- Ions, being positively charged and heavy, can be manipulated by magnetic and electrical fields.
- The ionization rate (or the measured ion current) is usually proportional to the gas density, the base for the ion gauges.
- An atom has a probability of being ionized that is dependent on the atom itself and the energy of the colliding electron. Thus the ion gauges are gas type dependent.
Electron Impact Ionization

- Electron impact ionization rate peaks at electron kinetic energy 50~200 eV for more gases.
- For hot filament gauges, electrons are emitted thermionically, and accelerated by an electric field.
- For cold cathode gauges, electrons are initiated by field-emission (or radiations), then trapped/amplified in a cross-field (electric and magnetic fields)

http://physics.nist.gov/PhysRefData/Ionization/Xsection.html
Hot Cathode Ionization Gauge – Principle

- Hot filament (cathode) emits electrons.

- Electrons collide with molecules and create positive ions.

- The positive ions that are created inside the grid volume are attracted to the collector and measured as ion current.

- The gauge controller electronics converts the collector ion current to a pressure reading.
HC Ionization Gauge – Basic Parameters

- Ionization cross section of gas molecule (size of molecule)
- Number of gas molecules present
- Number of ionizing electrons produced (emission current)
- Length of electron path (it is desirable to have the majority of the electron path inside the grid volume)
- Size of ionization (grid) volume
HC Ionization Gauge – Sensitivity

- For an electron beam with a path length $L$, the ionization yield (ions generated per electron) is:

$$nL\sigma_i = \frac{\sigma_i L}{kT} P$$

$\sigma_i$ – ionization cross section
$n = P/kT$ is molecular density

- If the electron current (emission current) is $i_e$, the total ion current:

$$i_i = \frac{\sigma_i L}{kT} \cdot i_e \cdot P = K \cdot i_e \cdot P = S \cdot P$$

$K = \frac{\sigma_i L}{kT}$ known as gauge coefficient

$S = K \cdot i_e$ is known as gauge sensitivity
Typical HC Ionization Gauge Sensitivity
HC Ionization Gauge – Relative Sensitivity

<table>
<thead>
<tr>
<th>Gas</th>
<th>Sensitivity</th>
</tr>
</thead>
<tbody>
<tr>
<td>Ar</td>
<td>1.2</td>
</tr>
<tr>
<td>CO</td>
<td>1.0-1.1</td>
</tr>
<tr>
<td>H₂</td>
<td>0.40-0.55</td>
</tr>
<tr>
<td>He</td>
<td>0.16</td>
</tr>
<tr>
<td>H₂O</td>
<td>0.9-1.0</td>
</tr>
<tr>
<td>N₂</td>
<td>1.0</td>
</tr>
<tr>
<td>Ne</td>
<td>0.25</td>
</tr>
<tr>
<td>O₂</td>
<td>0.8-0.9</td>
</tr>
<tr>
<td>Organic Solvents</td>
<td>&gt;&gt;1</td>
</tr>
</tbody>
</table>

\[ P_{\text{Gas}} = \frac{P_{N_2}^{\text{Gauge}}}{S_{\text{Gas}}} \]
For all HC ion gauges, the detected ion current always consists of a pressure dependent value, and a residual signal \( i_r \) that is not related to gas pressure.

\[
i_i = K \cdot i_e \cdot P + i_r
\]

The residual signal \( i_r \) sets the lowest measurable pressure by a HC gauge.

There are two major sources of the residual current:

- Soft X-ray induced current
- Electron Stimulated Desorption
**HC Ionization Gauge – Soft X-Ray Limit**

- Some electrons emitted from the hot cathode impact the grid and produce x-rays.
- Some of the x-rays impact the collector and produce photoelectrons.
- The exiting photoelectrons simulate positive ions arriving at the collector.
- The photoelectron current adds to the ion current producing an error in the pressure reading.
- Historic triode vacuum gauges had X-ray limit of $10^{-7}$ Pa. Modern HC gauges use much smaller anode to lower the limit below $10^{-9}$ Pa.

See Lafferty book P416 Fig 6.30 for triode HCG schematic
Electron Stimulated Desorption in HC Gauges

- Gas molecules are adsorbed on the surface of the grid.

- Electrons emitted from the cathode strike the grid and desorb the gas molecules.

- The electrons also ionize some of the gas molecules on the grid when they are desorbed.

- The additional gas molecules and positive ions contribute to an increase in the gauge pressure reading.
Ion Gauge Degas – Reduce ESD

**Resistive Heating**
- Grid Current: 7 A
- Grid Voltage: 8 VDC
- Filament
- Electron Bombardment
- Grid Voltage: 500 VDC
- Filament Emission Current: 100 mA
- Gas pumped away

**Electron Bombardment**
- Grid Voltage: 500 VDC
- Filament Emission Current: 100 mA
- Gas pumped away
Filament Selection

- **Thorium-coated Iridium**
  - General purpose
  - Operates cooler (~900°C)
  - Burn-out resistant

- **Tungsten**
  - Special purpose
  - Operates hotter (~1200°C)
  - Burns out easily and oxidizes when exposed to atmosphere
Bayard-Alpert Gauges

- Bayard-Alpert gauge (BAG) is mostly used in high- to ultra-high vacuum ranges ($10^{-4} \sim 10^{-11}$ torr), particularly the nude style.

- Typical BAG sensitivity for $N_2$ is 5~10/Torr.

- The small diameter center ion collector reduces the X-ray limit to low $10^{-11}$ torr.

- The caged grid can be degassed by heating to reduce ESD.

- The BAGs are robust and reliable.
**STABIL-ION® BAG Design**

Rugged Steel Enclosure

- Guard
- Port Shield
- Tensioned Filaments
- Precision-wound Grid
- Self-aligning Connector

*Brooks Automation - Granville Philips*
STABIL-ION® Gauge Types

- **Extended Range Gauge**
  - $1 \times 10^{-9}$ to $2 \times 10^{-2}$ Torr
  - x ray limit: $< 2 \times 10^{-10}$ Torr
  - Highest accuracy & stability
  - Sensitivity: 50/Torr

- **UHV Gauge**
  - $10^{-11}$ to $10^{-3}$ Torr
  - x ray limit: $<2 \times 10^{-11}$ Torr
  - Less accurate & stable than Extended Range Gauge
  - Sensitivity: 20/Torr

STABIL-ION gauges demonstrated excellent long-term reproducibility, many labs use them as reference calibration.
**MICRO-ION™ Gauge Design**

- **X-ray limit**: $< 3 \times 10^{-10}$ Torr
- **Upper pressure limit**: $5 \times 10^{-2}$ Torr/mbar.
- **Very compact, and low power.**
- **Good overlap with low vacuum** (> $1 \times 10^{-3}$ Torr) gauges such as **CONVECTRON ®**.
Deep UHV Gauge - Extractor

- Most widely used commercial XHV gauge.
- X-ray limit: $< 1 \times 10^{-12}$ Torr as the ion collector is recessed.
- Discriminate against ESD ions.
- Has the other features of a BAG, robust, replaceable filament and can be degassed.
- Range: $10^{-4}$ to $10^{-12}$ torr

LeyBold IE 511 Extractor Gauge
XHV Gauges - Energy Analyzers

Helmer Gauge
90° Bend Ion Analyzer

Bessel Box
Sold as Axtran® by ULVAC
Bent belt-beam gauge

- **X-ray limit:** $< 4 \times 10^{-14}$ Torr; $S_{N_2} = 2.8 \times 10^{-4}$/Torr.
- **Completely blocks ESD ions.**
- **Use the same controller as Extractor (IE511)**

*Fumio Watanabe, J. Vac. Sci. Technol. A 28(3) 2010, p.486*
In CCGs, a electron 'cloud' is created/trapped in a cross-field volume. Electrons gain energy through cyclic motions in the cross-field.

CCGs are gas-dependent in a similar way as HCGs.

The earliest CCG is a Penning ionization gauge.

In a CCG, the ion current is related to pressure as:

\[i_g = K \cdot P^n\]

\[n = 1.0 \sim 1.4\]
MKS-HPS Inverted Magnetron CCG

- Measuring range: $10^{-3}$ to $10^{-11}$ Torr (with Series 937 Controller)
- No X-ray limit
- Very low power, and no heating
- Very robust design
- Sensitive to contaminations, no degas option
- Be ware stray B-field
MKS-HPS Inverted Magnetron CCG (2)

\[ i_g = K \cdot P^n \]

- \( n = 1.09 \)
- \( n = 1.4 \)

CCG Circuit of operation
## HCGs versus CCGs

Both HCGs and CCGs are variable gauges in the range of $10^{-4}$ to $10^{-11}$ torr

<table>
<thead>
<tr>
<th></th>
<th>HCGs</th>
<th>CCGs</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Pros</strong></td>
<td>✓ Linear gauge response ✓ Higher gauge sensitivity ✓ Possible extension to XHV</td>
<td>✓ Inherent rugged ✓ Very low residual ion current ✓ Low power and heating ✓ Very good long-term reliability</td>
</tr>
<tr>
<td><strong>Cons</strong></td>
<td>✓ Higher X-ray and ESD limits ✓ Filament lifetime ✓ High power and heating ✓ Filament light</td>
<td>✓ Sensitive to contamination (oil, dielectric particulates, etc.) ✓ Discontinuity and nonlinearity ✓ Long ignition time at UHV ✓ Stray magnetic field</td>
</tr>
</tbody>
</table>
Partial Pressure Measurement

Residual Gas Analyzers
Why Residual Gas Analyzers

- All the gauges discussed earlier measures total gas pressure or density, no information on the gas composition.

- Residual gas analyzers are usually incorporated into critical vacuum system as vacuum diagnostic instrument.

- In most cases, qualitative mass spectral analysis is sufficient. Sometimes quantitative analysis is need, but rather difficult.

- A RGA measures relative signals verse mass-to-charge ratio (m/e), often in unit of AMU (atomic mass unit). (AMU is defined by $^{12}\text{C}$, that is, $^{12}\text{C}$ has exact 12.0000 AMU)
Components in a RGA System

- Control Electronics
- Ionizer
- Ion Filter
- Ion Detector
- Control & DAQ Software
Ionizers - Types and Parameters

- The most common ionizers are open ionizer, which is directly open to the vacuum to be monitored.
- Neutral gas molecules are ionized and fragmented by impact of electrons emitted from a hot filament.
- Ions are extracted and focused into a mass filter by a set of electrostatic lenses.
- Important ionizer parameters:  
  - Electron emission current
  - Electron energy
  - Ion energy
Electron Impact Ionization and Fragmentation

**Total Ionization 'Cross-Section'**

- **Fragmentation**
  - \( \text{CO}_2 + e \rightarrow \text{CO}_2^+ + 2e \)
  - \( \rightarrow \text{CO}^+ + \text{O} + 2e \)
  - \( \rightarrow \text{C}^+ + 2\text{O} (\text{O}_2) + 2e \)
  - \( \rightarrow \text{O}^+ + \text{CO} + 2e \)
  - \( \rightarrow \text{CO}_2^{++} + 3e \)
  - \( \rightarrow \text{CO}^{++} + \text{O} + 3e \)
  - \( \ldots, \ldots \)

Ions per centimeter electron path length per mbar at 20°C versus energy of incident electrons for various gases
Three Types of Ion Filters

- **Magnetic Sector**
  *Used mostly in leak checkers, large analytical mass spectrometers*

- **Quadrupole**
  *Most widely used in RGAs*

- **Auto-Resonant Trap**
  *Relatively new, only one manufacturer*
Magnetic Sector Ion Filter - Principle

- Ion Energy $V_0$ in eV
- Dipole field $B$ in Tesla

$$ r = \frac{1.44 \times 10^{-4}}{B} \left( \frac{MV_0}{z} \right)^{1/2} $$

$M$ - mass in atomic units
$z$ - degree of ionization
Magnetic Sector - Voltage Sweeping Mode

- Use permanent magnet, varying ion accelerating voltage $V_0$.
- Non-uniform $M/z$ scan with linear voltage sweep.
Magnetic Sector - Magnet Sweeping Mode

- Use electromagnet sector, and varying the B field.
- Uniform M/z scan with linear B field sweep.
Magnetic Sector Spectrometer Station

JEOL JMS-700 MStation

Resolution = 70,000 (10% Valley)
CDCl$_3$ (83.951808)
CH$_3$Cl$_2$ (83.953356)
M/ΔM = 54.260

Glucagon C$_{36}$H$_{55}$O$_{28}$N$_{11}$-H$_2$O (MW=1071.9528 g/mol)
Accelerating voltage: 7.0kV

M+H$^+$ 3481.6
Quadrupole Ion Filter - Principle

- Quadrupole Field: \( \Phi = (U + V \cos \omega t) \cdot \left( \frac{x^2 - y^2}{r_0^2} \right) \)
- The motion of ions in the quadrupole field can be solved using Mathieu's differential equations.
- Ions with certain M/z have stable trajectories to passing through exit aperture at given combination of U and V values.
### Quadrupole Ion Filter - A Non-Math Model

<table>
<thead>
<tr>
<th>xz plane</th>
<th>yz plane</th>
</tr>
</thead>
<tbody>
<tr>
<td>+</td>
<td>+</td>
</tr>
<tr>
<td>Rod: +U</td>
<td>Rod: -U</td>
</tr>
<tr>
<td>Transmission: full</td>
<td>Transmission: none</td>
</tr>
<tr>
<td>1</td>
<td>2</td>
</tr>
<tr>
<td>+</td>
<td>+</td>
</tr>
<tr>
<td>Rod: +U+V, cos ω</td>
<td>Rod: -U-V·cos ω</td>
</tr>
<tr>
<td>Transmission: low-pass</td>
<td>Transmission: high-pass</td>
</tr>
<tr>
<td>3</td>
<td>4</td>
</tr>
</tbody>
</table>

**Superimposition of the xy and yz planes**

#### DC field stabilize ions in one plane (XZ), deflect in the other (YZ)

**Superimposed RF field 'kicks' lighter ions in one plane (XZ), while 'corrects' heavy ions in the other (YZ)**

**Combination of both 'low-pass' and the 'high-pass' to form a 'band-pass' ion filter**
Quadrupole Ion Filter – Characteristics

- Linear mass scan (uniform mass separations) with constant peak width can be achieved for most commercial MQS, when properly tuned.

- Ion filter transmission usually depends on M/z. In many instruments, the transmission efficiency (TF) decreases in high mass (>20): TF \propto 1/(M/z). Transmission factor also depends on ion energy.

- Tuning and calibration is usually more difficult, must be done by experts, or in factory. In most brands, the manufacturers strongly recommend one-to-one match of the control electronic unit and the sensor head.
**Auto-Resonant Trap – Principle**

- **Electrostatic Ion Trap**: Ions confined by purely electrostatic fields oscillate at a resonant frequency inversely proportional to \( \sqrt{m/z} \)

  Where, \( m \) is mass, \( z \) is the total charge of the ion

- **Autoresonance**: RF scan pushes ions when scan frequency matches ion’s resonant frequency

**Electrostatic confinement = Ultra-low power requirements**
Auto-Resonant Trap – Characteristics

- Very fast scans (as fast as 85 ms/scan to 300 AMU)
- Very compact, low power and low RF power
- Much less artifact peaks, with very low electron emission current (as low as 5-μA)
- ART is ratio-metric. Need a total pressure gauge for ‘true’ partial pressure measurement
- High background ion current at high pressure (>10⁻⁷ torr)
Ion Detectors - Faraday Cup

- At high-vacuum ($10^{-5}$ to $10^{-8}$ torr), a Faraday cup style charge detector is sufficient.
At UHV conditions, the ion current becomes too small to be directly measured by a Faraday cup, electron multipliers are used.

The electron multipliers are relying on secondary emission process on active coatings.

To achieve sufficiently high gain, usually multiple stages of secondary emissions are employed.

Though discrete dynodes yield very high gain, continuous dynode multipliers are most commonly used in RGAs, as their active coatings are less sensitive to air exposure (oxidization degradation).

Two types continuous dynode multipliers: Micro-Channel Plates and Channeltrons
Ion Detectors - Channel Plate

At UHV/XHV (<10^{-8} \text{ torr}), a electron multiplier is needed. Micro-channel plate (MCP) is a lower cost multiplier with gain up to 10^5.
At UHV/XHV (<10^{-8} torr), a electron multiplier is needed. Channeltron® has much higher gains (>10^7) at higher cost.
RGA - Operational Parameters

- **Mass range** - For most HV, UHV, XHV systems, 0-100 AMU is sufficient.

- **Resolution** - Normally, RGA’s resolution $\Delta M$ is set ~1.0. Lower resolving power ($\Delta M>1$) may be needed to gain sensitivity.

- **Signal sensitivity** - Most modern RGAs claim overall sensitivities $10^{-14}$ torr. The RGA sensitivity depends on the Faraday-cup sensitivity (RGA’s basic sensitivity), and the gain of electron multiplier.

- **Mass scan speed** - Most quadrupole RGAs can scan 0-100 AMU in seconds, ART-MS in <100 ms
Some Commercial RGA Systems

**Inficon Transpector 2**

- **Range:** 0 -100, 0 -200, 0 -300
- **Pressure range:** <10^{-4} torr
- **Sensitivity** (amp/torr):
  - 10^{-4} (FC), 500 (EMP)
- **Minimum detectable PP** (torr):
  - 3x10^{-13} (FC)
  - 5x10^{-15} (EMP)

**Brooks Automation VQM**

- **Range:** 0 -145, 0 -300
- **Pressure range:** <10^{-5} torr
- **Sensitivity** (amp/torr):
  - Depend on total pressure
- **Minimum detectable PP** (torr):
  - Depend on total pressure
  - (<10^{-13} torr) (always require EMP).
MKS’s Micro-Vision Quadrupole RGA with bakeable and radiation resistance cable allows continuous operations in the accelerators. Many (100s) are used at Diamond Light Source, some has been in use in CESR for over 5 years, with no radiation induced damage, however, with much higher price tags.
Cracking Patterns

- When molecules of a gas or vapor are struck by electrons with sufficient kinetic energy, ionization and fragmentation of the molecules may occur, resulting in ions with several mass-to-charge ratios.

- The mass-to-charge values are unique for each gas species, with a distribution (or pattern) of relative intensity of these M/z peaks, depending on the gas species (and the instrumental conditions).

- The distribution or the pattern is often referred as **cracking pattern** of the gas species.

- Besides singly ionization of a molecules (\( \text{CH}_4 \rightarrow \text{CH}_4^+ \)), at least two more factors contributed to the cracking pattern: dissociative ionization (fragmentation) and isotopes.
Dissociative ionization – CH₄ as example

![Graph showing dissociative ionization of CH₄](image)
Isotope Effect - Ar as example
Combined Effect - CO as example

Combined Effect - CO as example

Relative Intensity

M/z

$^{12}C$, $^{13}C$, $^{16}O$, $^{12}C_2$, $^{12}C_2$, $^{12}C_2$, $^{12}C_2$
Cracking patterns are commonly used as “fingerprints” of a gas or a vapor, for qualitative gas analysis.

Cracking patterns of many common gases and vapors can be found in the literatures.

Published cracking patterns should be used as a guidance, and they not only depend on gas/vapor, but also vary with instrument conditions.

Many commercial RGA systems have ‘build-in’ gas library. NIST also maintain a online mass spectrum data.

http://webbook.nist.gov/chemistry/
Qualitative Analysis - Example 1

- An unbaked vacuum system, $\text{H}_2\text{O}$ dominant
- An clean UHV chamber, $\text{H}_2$ dominant
Vacuum system with a small air-to-vacuum leak!

Relative Intensity

M/z (AMU)

Pressure Torr

Relative Intensity
Qualitative Analysis - Example 3

A vacuum system contaminated by long-chain hydrocarbons, with peak-grouped by AMU=14, that is fragmenting by breaking a CH$_2$ species.
RGA - Quantitative Analysis

- Quantitative analysis of partial pressures is NOT necessary in most cases, as it is very difficult and often inaccurate.

- For a system contains gases with non-overlapping peaks are relatively straightforward, by calculating PP of individual gases, using a dominant peak. Example: CO (amu:28), Ar (amu:40).

Calculating CO partial pressure using ion current at M/z=28

\[
PP_{CO} = \frac{I_{CO_{28}}}{FF_{CO_{28}} \cdot XF_{CO_{28}} \cdot TF_{28} \cdot DF_{CO_{28}} \cdot G \cdot S}
\]

Calculating Ar partial pressure using ion current at M/z=40

\[
PP_{Ar} = \frac{I_{Ar_{40}}}{FF_{Ar_{40}} \cdot XF_{Ar_{40}} \cdot TF_{40} \cdot DF_{Ar_{40}} \cdot G \cdot S}
\]

\[FF_{(CO_{28}, Ar_{40})}: \text{Fragmentation ratio of (CO, Ar) to M/z=(28,40) - Cracking Pattern}\]

\[XF_{(CO_{28}, Ar_{40})}: \text{Relative ionization probability (to N}_2\text{) of (CO, Ar) to (CO}^+\text{, Ar}^+\text{) ions}\]

\[TF_{(28, 40)}: \text{Transmission factors for M/z=(28, 40) ions through mass filter}\]

\[DF_{(CO_{28}, Ar_{40})}: \text{Detector factor of (CO}^+\text{, Ar}^+\text{) ions}\]

\[G: \text{multiplier gain}; \quad S: \text{Instrument sensitivity for N}_2\text{, in Amp/Torr}\]

Gauge Selection Considerations

- **Gauge Range**: Multiple gauges should be installed in an accelerator vacuum system to cover pressure ranges from atmospheric pressure to the working pressure.
  - Convectron Pirani gauges are ideal for pressure atm. $\sim 10^{-3}$ torr
  - Ionization gauges are usually used for HV/UHV ranges.

- There are commercial 'full-range' combination gauges from many vendors.

- RGAs should be installed for UHV accelerator vacuum systems to monitor vacuum system performances and online trouble-shooting, such as potential leaks during operations, and/or unusual beam induced pressure rises.
Gauge Application to Accelerators

- Gauges’ long-term (multi-year) reliability is most important factor, as in many accelerator vacuum system, access to the gauge heads can be very limited. In this aspect, CCGs are preferred over HCGs.

- Gauges with ‘on-board’ electronics should always be avoided, due to the radiation damage. One should also consider the long cable factor for selecting a type of gauges. Gauges require low power is always a plus.

- When installing a gauge to the accelerator beam pipe, one should avoid line-of-sight of the gauge port to the particle beam, to minimize the ‘cross-talks’ between the gauge and the beam.
# Accelerator Vacuum Gauges - A Survey

<table>
<thead>
<tr>
<th>Institute Accelerator</th>
<th>Total Pressure Gauge</th>
<th>Partial Pressure Gauge</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Type/Brand</td>
<td>Q’ty</td>
</tr>
<tr>
<td>Cornell / CESR</td>
<td>CCG / MKS</td>
<td>~90</td>
</tr>
<tr>
<td>ANL / APS</td>
<td>CCG / MKS</td>
<td>~140</td>
</tr>
<tr>
<td>BNL / RHIC</td>
<td>CCG / MKS</td>
<td>&gt;100</td>
</tr>
<tr>
<td>BNL / NSLS II</td>
<td>CCG / MKS</td>
<td></td>
</tr>
<tr>
<td>FNL / Tevtron</td>
<td>HCG/CCG</td>
<td>~100/40</td>
</tr>
<tr>
<td>CERN / LHC</td>
<td>BAG / CCG</td>
<td>~250 BAG</td>
</tr>
<tr>
<td>KEK SuperB</td>
<td>CCG / DIA VAC</td>
<td>~300</td>
</tr>
<tr>
<td>ESRF</td>
<td>CCG / Pfeiffer</td>
<td>~200</td>
</tr>
<tr>
<td>JLAB</td>
<td>HCG (EXT) for ERL-FEL CCGs for CEBAF</td>
<td>100s</td>
</tr>
<tr>
<td>TLS – TPS</td>
<td>Extractor / Leybold</td>
<td>~280</td>
</tr>
<tr>
<td>Diamond L.S.</td>
<td>CCG / MKS</td>
<td></td>
</tr>
<tr>
<td>Pohang L.S.</td>
<td>BAG&amp;CCG (Pfeiffer)</td>
<td>~80</td>
</tr>
</tbody>
</table>