

Vacuum Science and Technology for Accelerator Vacuum Systems

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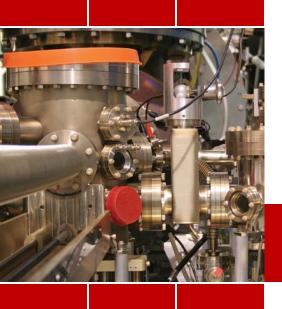


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SESSION 5.4: FABRICATION AND CLEANING

- ☐ Cleanness is the foundation in achieving UHV and required vacuum system performance.
- ☐ Cleanness starts with vacuum component material selection, and fabrication.
- □ Proper fabrication processes leads to easier and less aggressive post-machining cleaning.
- ☐ Cleanness requires a development of UHV practice and culture, besides the adequate facilities.
- □ UHV clean and Particle-free are very different requirements, though most modern accelerator vacuum systems demanding both.

Material Preparation Techniques



- Vacuum materials may be prepared for finish machining by the following techniques:
 - 1. Rough Machining
 - 2. Metal Stamping
 - 3. Water-jet cutting
 - 4. Laser cutting
 - 5. Plasma arc cutting
 - 6. Bead/sand blasting
- When plasma arc cutting, make sure that sufficient material allowance is made for complete removal of the heat affected zone (HAZ) during final machining.
- Bead/sand blasting should only be permitted on material with large amounts of mill scale or heavy inclusions from contact with metallic or organic material.





Material Finishing Techniques



- The preferred technique for finishing vacuum materials is machining (turning and milling).
- The following techniques should be avoided or at least approved on a case-by-case basis:
 - 1. Grinding
 - 2. Honing
 - 3. Electric Discharge Machining (EDM)
 - 4. Chemical milling
 - 5. Glass bead blasting
- EDM (specially wire-EDM) is acceptable when DI-water is used as dielectric fluid. Otherwise more aggressive etching may be necessary.
- Glass bead blasting may be permitted with new clean beads when an optically dispersive surface is required. (Alumina beads are better choice.)





Material Finishing Techniques (cont.)



- When machining will not produce the required surface finish, polishing may be permitted. When polishing, care should be taken to avoid excessive rubbing or contact pressure.
- > The following abrasives are acceptable for UHV components.
 - 3M Scotch Type S, Silicon Carbide (color: gray), 500 grit
 - Brite Type A, Aluminum Oxide (color: maroon), 240 grit
 - 3M Wet or Dry Fabricut Cloth Aluminum oxide or silicon carbide, 600 grit
 - Mineral oil suspended diamond pastes
- > Modern extrusions with proper dies can achieve almost "mirror-like" inner surface finishes.
- Very smooth inner surface finish via a process called "Abrasive Flow machining, or <u>AFM</u>". This process was employed for the LCLS's long undulator beam pipes.





Cutting Fluids for Finish Machining



If feasible, only alkali-detergent like cutting fluids should be used for the finish machining. This type of oil-free fluids are fully water soluble, and are readily removed by solvents. Some example fluids are:

	<u> </u>
Relton A-9	Rapid Tap
Tap Magic	Trim Tap
Tapmatic #1 or #2	RD2-195
"Pearl" Kerosene by Chevron Chem CO	Dip Kool 868
"Tool Saver" by Do All Corp.	DIP Kool 862
Cutzol EDM 220-30	Dip Kut 819H
Sunnen Man-852 Honing Oil	No Sul #6871
Vytron Concentrate	Kool Mist #88
Rust-Lick G-25-J	Cimcool 5 Star 40
Wheelmate #203	Cimperial # 1011
Aqua Syn 55 by G-C Lubricants CO	Haloform CW-40
Cold Stream Coolant by Johnson Wax CO	Trim Sol
"Acculube" by Lubricating Systems Inc.	<i>Trim9106C5</i>
Micro Drop "Advanced System Lubricant" by Trico	CINDOL 3102
Micro Drop "New Vegetable Based" by Trico	PenWalt #DP 1131

Outgassing (with RGA) tests of machined specimen (after cleaning) may be necessary to establish acceptability of a machining fluid.





Dry-Machining or Solvent as Cutting Fluid



- ☐ In reality, machining on a finished chamber is needed. In those cases, dry-machining is required.
- ☐ In some facilities, such as during TPS's (Taiwan Light Source) aluminum chamber fabrication, machining in a Clean Room using ethanol as cutting fluid has been implemented.



Oil-free CNC Machining in Clean Room

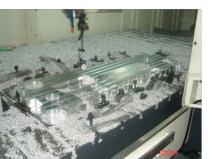




Tool enter increment: $0.3 \sim 0.5$ mm



Tool enter increment: $0.1 \sim 0.2$ mm



Tool enter increment: < 0.05mm



CNC Machining Processes (with 99.5% Ethanol as lubricant)





Suggested UHV Handling & Assembly Guidelines



- · No food, drink, or smoking allowed in CLEAN AREA.
- · Limit entry and exit into CLEAN AREA.
- Hydrocarbons (oils, grease) and dust-collecting materials (cardboard) must be minimized.
- Equipment brought into CLEAN AREA must be clean. Carts, chambers, stands, and tools must be free of oils and dust.
- · Wood must be minimized. If wood is required, it should be wrapped in "ULO" plastic sheet and/or cleanroom tape.
- · A special set of tools expressly for use on vacuum components should be kept in the CLEAN AREA.







- · Metal tools must be degreased. After degreasing, tools should be kept in clean trays and handled with clean gloves.
- · No cadmium plated, lead, or painted tools should be permitted. Chrome and nickel plated tools are permitted.
- · Aluminum foil shall be in accordance with ASTM B479, type designated as DRY ANNEAL A, (oil free). Each piece of foil should be used only once and then discarded.
- · Aluminum foil and lint-free tissue should be stored in clean boxes with lids.
- · Only use pens for writing in CLEAN AREA, do not use pencils.

 Minimize the use of paper. If paper is used, it should be "cleanroom type."







- · Clean vacuum parts and open chambers should be covered with foil at all times when work is not being performed.
- Do not wear wooly sweaters in CLEAN AREA.
- No sandpaper or abrasives allowed.
- · Hands should be kept out of pockets (this introduces lint).
- · Clean parts should be handled with new polyethylene gloves used inside 100% stretch nylon gloves.
- · Gloved hands which touch cleaned parts and tools should touch nothing else (this includes your face, hair, etc.). Gloves which touch unclean surfaces should be replaced immediately.







- Replace gloves with a new, clean pair at the beginning of each shift and following breaks.
- Hands should be washed before wearing clean gloves.
- Clean-room quality protective clothing (lab coats, hats, hair nets, face masks) should be worn when working on vacuum components in CLEAN AREA.









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A Generic Cleaning Procedure



Mechanical Cleaning



Detergent Cleaning*

Chemical Etch*

Electrolytic Polishing*

High Pressure Rinse

Degassing



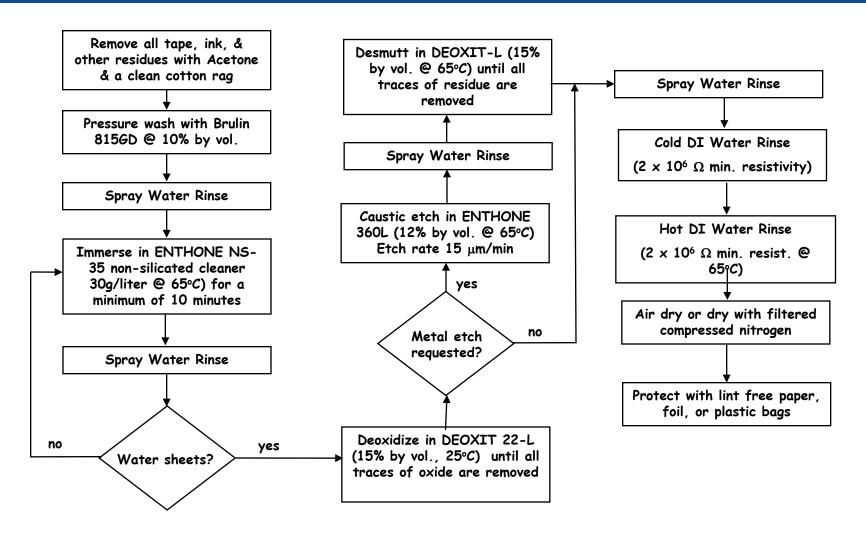
*Typically these steps are proceeded by a water rinse to avoid contamination of subsequent baths.





Cleaning of Aluminum Components





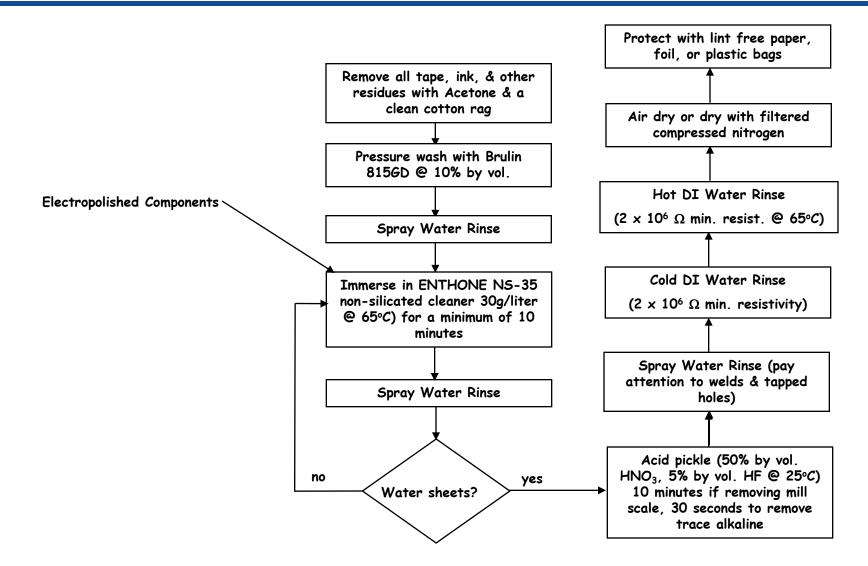
Ref. LLNL MMED Plating Shop Procedure





Cleaning of Stainless Steel Components





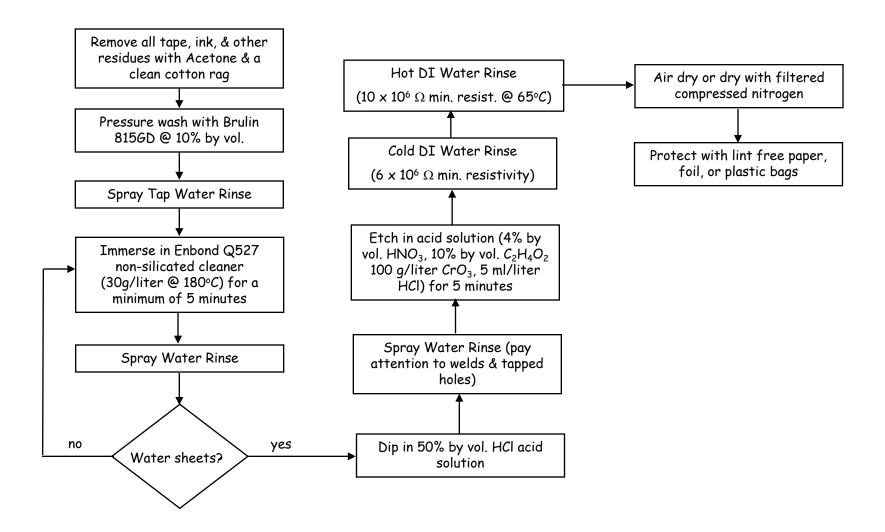
Ref. LLNL MMED Plating Shop Procedure





Cleaning of Copper and Glidcop Components





Ref. LLNL MMED Plating Shop Procedure





Cleaning Procedure Varies with Labs



CLASSE, Cornell Vacuum Part Cleaning Request Card

CHEM. ROOM JOB TICKET								
]	Part name: These basic cleaningAcent #:							
	Submitted by:	Date submitted:						
]	Date required:	usually sufficient	<u>Material:</u>					
			Quantity:					
	🗖 Hand-scrub,	soap & water						
[🗖 Detergent & 1	ultrasonic						
	Solvent only in ultrasonic							
Chemical etch:								
	☐ S.S. cold polish ☐ Aluminum etch							
- {	Copper cleaner Citric acid							
	□ BCP 1:1:2	☐ Citranox & water						
Special instructions (times, temp., formulas, etc.):								
	These aggressive cleanings are used only							
	necessary, such as parts for vacuum braze							





Some of the More Aggressive Recipes



NEW Copper Cleaner:

OLD Copper Cleaner:

1450 M.L. D.I. H₂O 1750 M.L. HNO₃ 800 M.L. H₃PO₄ 1200 M.L. D.I. H₂O 1600 M.L. HNO₃ 800 M.L. H₃PO₄ 400 M.L. HF

Aluminum Etch:

sodium it x inxide

Dissolve 6 oz. NaOH in 4 liters Hot H2O

Desmut:

500 M.L. D.I. H2O 190 M.L. H2SO4=SylfuRic 5 oz. Deox #7 Chromate Then add 3300 M.L. D.I. H2O

S.S. Cold Polish

500 M.L. H₂O 716 M.L. HF 1000 M.L. HNO₃ Then add 1684 M.L. Hot H₂O

S.S. Pickle:

400 Pickle 1000 M.L. HCI 120 M.L. HNO₃ 2480 M.L. Hot H₂O

Ti Pickle:

80 M.L. HF 1200 M.L. HNO₃ 2720 M.L. D.I. H₂O

By using water-soluble cutting fluid, these more aggressive etching were used much less at CLASSE, Cornell.





Cleaning with Ozonate Water



- ☐ An effective cleaning method with ozone gas for the reduction of carbon contamination from the surface has been investigated by T. Momose [1].
- ☐ Another method of ozonate water cleaning for the superconducting RF cavities studied by K. Asano [2] shows good results of removing the surface contaminations.
- ☐ As a much more environmental friendly cleaner, ozonate water is also find wide-spread use in semiconductor industries for wafer cleaning.
- ☐ In TPS and NSLS II vacuum system fabrication, ozonate water cleaning has been employed as the primary cleaning for the aluminum chambers. (Data for Cu showed less benefit for ozonate water, as compared to Citranox® cleaning.)

[1] T. Momose, Y. Maeda, K. Asano and H. Ishimaru, J. Vac. Sci. Technol. A13(3), 515 (1995). [2] K. Asano, T. Furuya, S. Mitsunobu, T. Tajima and T. Takahashi, "Stable Performance of 508-MHz Superconducting RF Cavities for KEK B-Factory", KEK Preprint 95-191 (1996).





TPS's Ozonate Water Cleaning



Ozonated Water Cleaning (> 20 ppm)



3. Drying in Clean booth

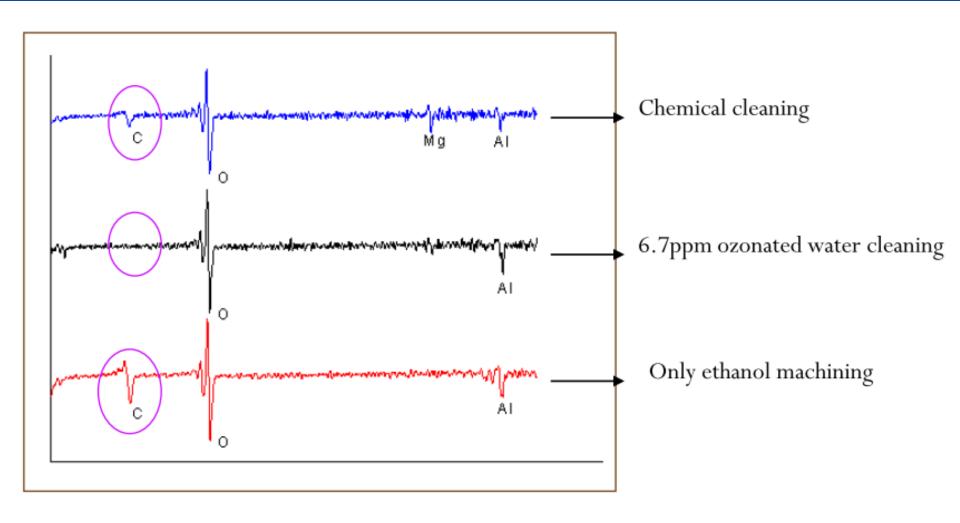
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Surface analysis (AES) for Al samples





G.Y. Hsiung

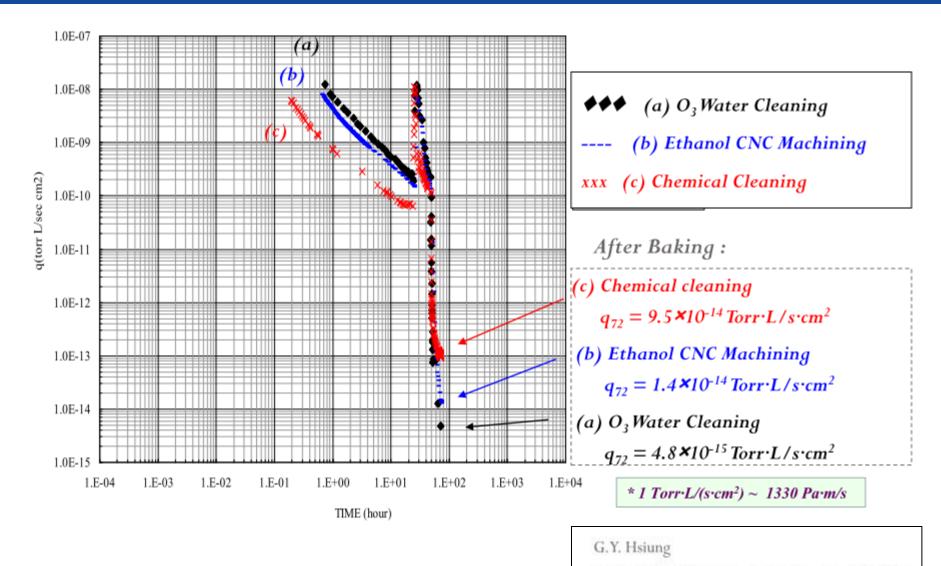
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Thermal Outgassing Rate (Aluminum)







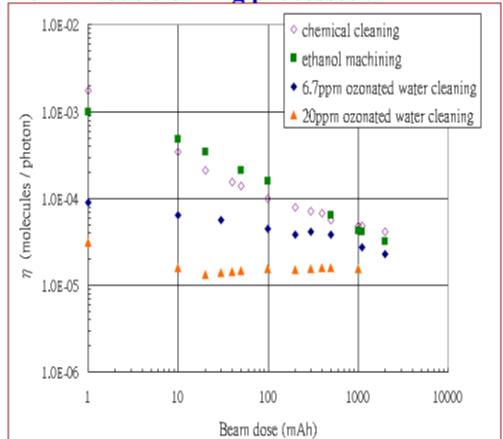


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PSD yield for Aluminum samples



Yield of photon stimulated desorption (η) for various cleaning processes :



- η of ozonate water cleaning is more than one order of magnitude lower than those of only ethanol machining or chemical cleaning at 1mAh.
- η decreases with beam dose increases.
- Al samples with ozonate water cleaning maintain the lowest η at the beam dose through 1000 mAh.

G.Y. Hsiung

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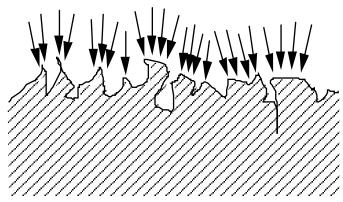


Electropolish

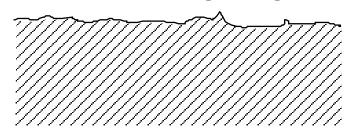


- Consider as a "reverse" electroplating technique.
- Metal is removed from the "high spots" due to higher current density.
- Surface metal is rich in H₂ and fluid until degassed.
- Electropolish produces a bright metallic finish.
- With proper rinsing and a post bake step, very low outgassing rates can be achieved.

Electric Field Lines are concentrated at peaks producing higher chemical milling rates



Resulting surface has reduced peaks, reduced surface area, and reduced outgassing





High Pressure Rinse (HPR)



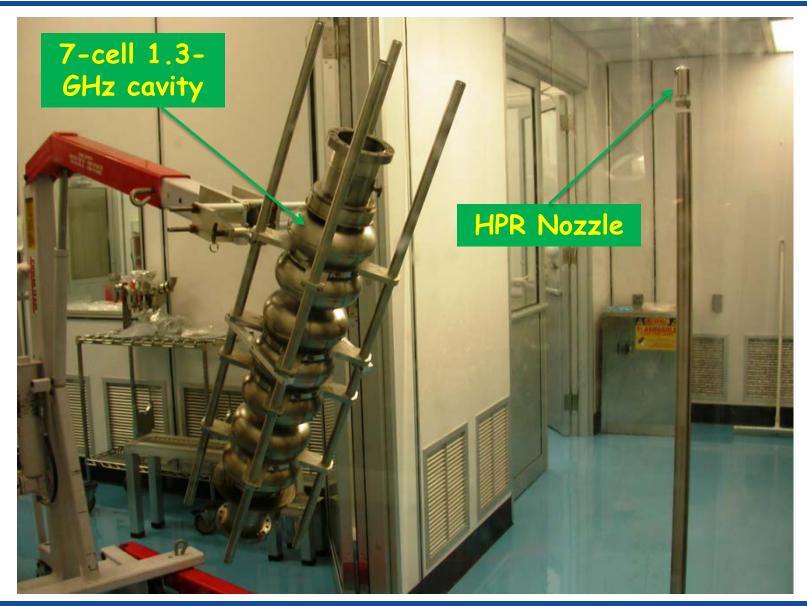
- · Fluid used can be tap water, deionized water, or with a detergent to assist in cleaning.
- · With a detergent, this process is used early in the cleaning process. With deionized water, it is one of the final steps.
- · Use of high fluid velocity to dislodge particles from the surface.
- · Most effective cleaning method for particles in the 1 μ m range.
- · High pressure rinse can be effective on large parts, as well as small parts.





HPR SRF Cavity - Final Cleaning 1









HPR SRF Cavity - Final Cleaning 2



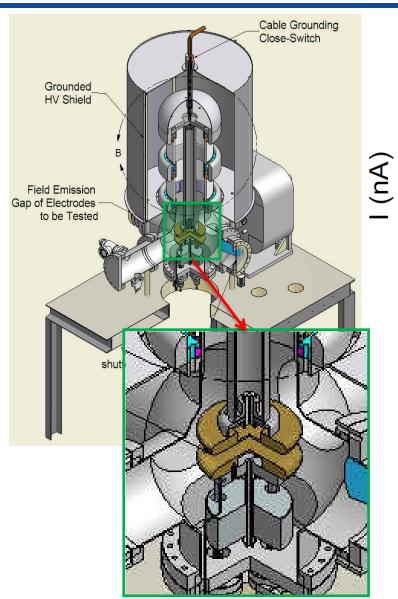


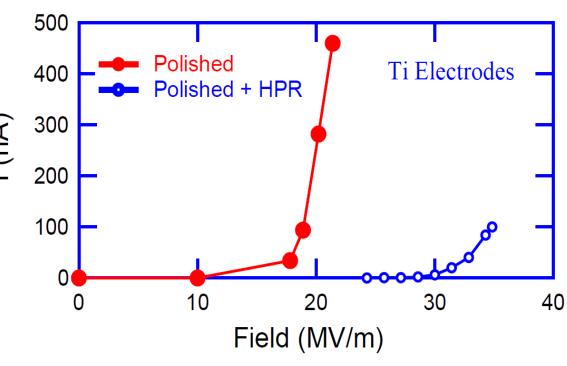




HPR HV Electrode for ERL







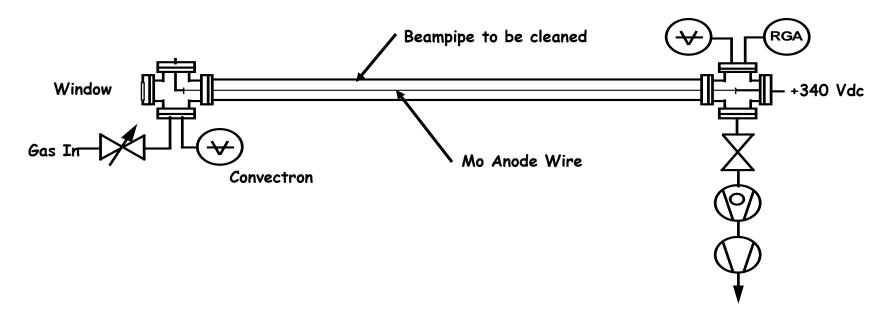
HPR approved to be very effective to remove field emitters for Cornell ERL DC gun structure





Glow Discharge Cleaning Schematic





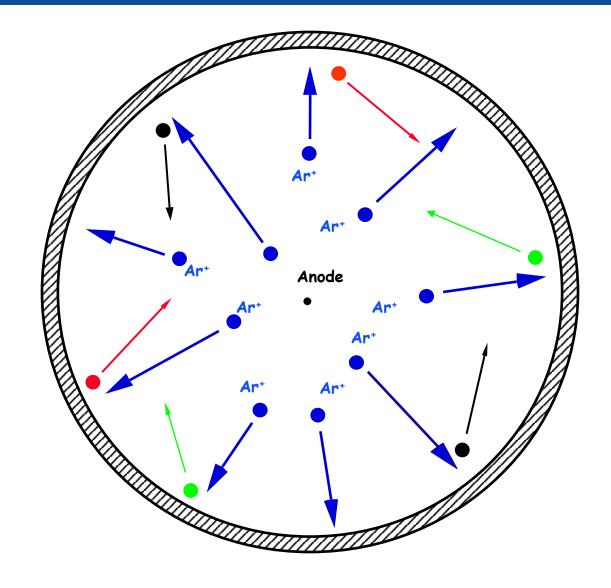
- · Glow discharge cleaning is useful in removing surface contamination such as C, S, H_2O , and organics.
- · Must be a flowing system to prevent readsorption.
- · Typical gases used are Ar, Ar- O_2 , H_2 .
- · Glow discharge cleaning can leave higher levels of Ar and O_2 in the metal surface.
- · A 200°C bakeout is still required after glow discharge cleaning.





Glow Discharge Cleaning









Various surface treatments evaluated by XPS



Surface analysis by x-ray photoelectron spectroscopy (XPS)

	XPS Surface Atom %					
Surface Treatment	Cu	0	N	С	Cl	Ar
Chem. Cleaning (old SLAC recipe)	22.4	22.5	11.9	41.6	1.6	-
Chem. Cleaning (new SLAC recipe)	43.4	36.8	-	17.9	1.9	-
GDC - 95% Ar, 5% O_2 (2 x 10^{19} ions/cm ²)	50.6	40.0	-	8.0	-	1.4
GDC - 95% Ar, 5% O_2 (2 x 10^{18} ions/cm ²)	48.6	42.0	-	8.0	-	1.4
GDC - $100\% H_2$ (2 x 10^{18} ions/cm ²)	64.2	23.6	-	12.2	-	-

Ref. "Processing of OFE Copper Beam Chambers for PEP-II High Energy Ring", Hoyt et al, 1995 Particle Accelerator Conference





Bakeout and Degassing



· Vacuum firing of components will result in low outgassing rates

 $(T = 800^{\circ}C - 1000^{\circ}C, P \sim 10^{-4} \text{ Torr for several hours}).$

- Bulk H_2 is depleted from metal
- Works well for stainless steels
- copper and aluminum are annealed
- · Heating systems for bakeout
 - Ovens are the easiest to use
 - Heater tapes with insulation
 - Nichrome wire covered with ceramic beads
 - Calrods or heater bands with insulation
 - Heater blankets (built-in insulation)





SLAC Glow Discharge and Bakeout Station





Glow Discharge Station

Bakeout Oven Enclosure (200°C)

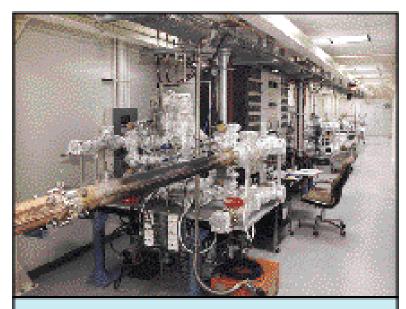




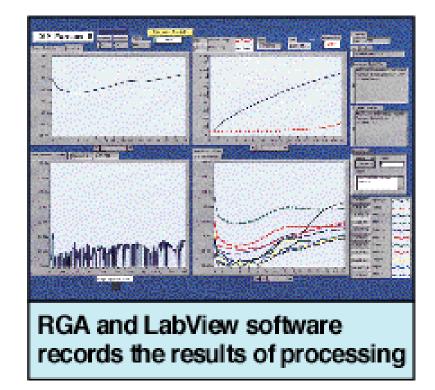


LLNL Glow Discharge and Bakeout Station





ATEG glow discharge and bake station







In-situ bakeout of Cornell ERL Beamline









Cornell's Recipe for Degassing SSTs



- The heat-treated (400°C bake in air for a duration of up to 100 hr) all stainless steel materials for the gun chamber.
- SST outgassing rates (after a 150°C vacuum bakeout) below 10^{-13} torr·l/s·cm² achieved with F_0 >3, a dimensionless time scale, F_0 .

$$F_0 = 4Dtd^{-2}$$

- t heating time; d -thickness
- D=D₀ exp(-E_d /kT) Hydrogen diffusion constant

