Vacuum Science and Technology for Accelerator Vacuum Systems

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Material Selection Considerations

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2. Chemical properties
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3. Mechanical properties
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4. Fabrication properties
   Machinability, formability, weldability, etc.

5. Vacuum properties
   Outgassing rate, porosity, bakeability, etc.

6. Surface modification and engineering
   Conductive thin films, functional coatings, etc.
Stainless Steels

- Stainless steels are most commonly used vacuum construction materials for accelerators, for their high strength and hardness, corrosion resistance, bakeable to high temperature (up to 450 °C under vacuum, degassing up to 900 °C), excellent weldability, excellent formability, etc.

- However, stainless steel’s low thermal conductivity may result in very high (thus unsafe) thermal stress if exposed to localized high heat flux (intense synchrotron radiation, etc.)

- Certain stainless steels (even austenitic alloys) may be magnetized from cold-world (bending, etc.) and/or welding.

- For high intensity, short bunched electron/positron storage rings, high electric resistivity of stainless steels may also have negative impact to the accelerator performance.

- Stainless steels may also become radioactive when bombarded by high energy particles.
Stainless Steels – Classifications

- Stainless steel is a steel alloy with at least 11% chromium content by weight. It is also called corrosion-resistant steel.

- Austenitic stainless steel (300 series): These are generally non-magnetic steel alloys. They contain a maximum of 0.15% carbon, a minimum of 16% chromium and sufficient nickel and/or manganese to retain an austenitic structure at all temperatures from the cryogenic region to the melting point of the alloy. The low carbon (-L) grades are used when welding is involved. In UHV applications, especially accelerators, 300 stainless steels are commonly used.

- Other less used types of stainless steel alloys are: Martensitic stainless steels (400 series), precipitation-hardening martensitic stainless steels (most common used 17-4PH) and ferritic stainless steels. Martensitic stainless steels are much more machinable, and are magnetic. Those are much less used for accelerators, mainly due to the magnetism.
Austenitic Stainless Steels

- High strength, moderate formability, excellent weldability.

- Can be extruded in simple shapes

- **304L SS**, most commonly used in vacuum, but may become magnetized from machining and welding.

- **316L SS**, with Mo added, more expensive, resistant to chemical attack, welds are non-magnetic

- **316LN SS**, a nitrogen-enhanced 316L steel, much more expensive, but excellent strength at very elevated temperatures (as high as 1000°C)

- Wide variety of circular tubes and pipes available (seamless & welded)

- Outgassing rates can be decreased by employing good machining techniques, chemical cleaning and baking (up to 900°C)

- Poor thermal and electrical conductivity
Family of Austenitic Stainless Steels

- **302B**
  - Sr added to increase scaling resistance

- **304**
  - General purpose

- **202**
  - N & Mn partially replaces Ni

- **205**
  - N & Mn partially replaces Ni

- **201**
  - N & Mn partially replaces Ni

- **302**
  - N increased to lower work hardening

- **304**
  - S added to improve machining

- **304L**
  - C reduced even further

- **316**
  - C reduced for welding

- **317L**
  - More Mo & Cr added for better corrosion resistance

- **316L**
  - C reduced for welding

- **310S**
  - Same as 309 only more so

- **347**
  - Ti added to prevent carbide precip.

- **321**
  - Cb added to prevent carbide precip.

- **314**
  - Si increased for highest heat resistance

- **348**
  - Ta & Co restricted for nuclear applications

- **304N**
  - N added to increase strength

- **30430**
  - Cu added to improve cold working

- **303Se**
  - Se added for better machined surfaces

- **384**
  - More Ni to lower work hardening

- **316F**
  - S & P increased to improve machining

- **317LMN**
  - Mo added N added

- **316N**
  - N added to increase strength
Stability of Austenitic Stainless Steels

- Cold-work (rolling, forging, bending, etc.) of austenitic stainless steel may induce transformation into martensitic phase (thus magnetic).

- Type 316 alloy (with Mo) is more stable than other 3xx type alloys.

- Magnetized stainless steel may be annealed back to austenitic phase, or ‘de-gaussed’ with strong alternating magnetic coil.
# Mechanical Properties for Stainless Steels

<table>
<thead>
<tr>
<th>Property</th>
<th>304L</th>
<th>316L</th>
<th>316LN</th>
<th>OFE Cu</th>
</tr>
</thead>
<tbody>
<tr>
<td>Ultimate Tensile Strength (MPa)</td>
<td>564</td>
<td>560</td>
<td>637</td>
<td>338</td>
</tr>
<tr>
<td>Tensile Strength (ksi)</td>
<td>81.8</td>
<td>81.2</td>
<td>92.4</td>
<td>49.0</td>
</tr>
<tr>
<td>Yield Strength (Mpa)</td>
<td>210</td>
<td>290</td>
<td>&gt;280</td>
<td>217</td>
</tr>
<tr>
<td>Yield Strength (ksi)</td>
<td>30.5</td>
<td>42.1</td>
<td>&gt;41.6</td>
<td>31.5</td>
</tr>
<tr>
<td>Elongation at Break (%)</td>
<td>58</td>
<td>50</td>
<td>58</td>
<td>55</td>
</tr>
<tr>
<td>Modulus of Elasticity (Mpa)</td>
<td>197</td>
<td>193</td>
<td>200</td>
<td>115</td>
</tr>
<tr>
<td>Modulus of Elasticity (ksi)</td>
<td>28.6</td>
<td>28.0</td>
<td>29.0</td>
<td>16.7</td>
</tr>
</tbody>
</table>

Ref. [www.matweb.com](http://www.matweb.com)
<table>
<thead>
<tr>
<th>Property</th>
<th>304L</th>
<th>316L</th>
<th>316LN</th>
<th>OFE Cu</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Composition:</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>C</td>
<td>0.03%</td>
<td>&lt;0.03%</td>
<td>&lt;0.03%</td>
<td>Cu 100%</td>
</tr>
<tr>
<td>Cr</td>
<td>18-20%</td>
<td>17.9%</td>
<td>17.9%</td>
<td></td>
</tr>
<tr>
<td>Mn</td>
<td>2%</td>
<td>2.0%</td>
<td>2.0%</td>
<td></td>
</tr>
<tr>
<td>Mo</td>
<td></td>
<td>2.5%</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Fe Balance</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Ni</td>
<td>8-12%</td>
<td>11-14%</td>
<td>10.8%</td>
<td></td>
</tr>
<tr>
<td>P</td>
<td>0.045%</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>S</td>
<td>0.03%</td>
<td>0.03%</td>
<td>0.16%</td>
<td></td>
</tr>
<tr>
<td>Si</td>
<td>1%</td>
<td>1%</td>
<td>1%</td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td><strong>Melting Point (°C)</strong></td>
<td>1427</td>
<td>1385</td>
<td>1400</td>
<td>1083</td>
</tr>
<tr>
<td><strong>Density (g/cc)</strong></td>
<td>8.0</td>
<td>8.0</td>
<td>8.0</td>
<td>8.92</td>
</tr>
<tr>
<td><strong>Electrical Resistivity (Ω-cm)</strong></td>
<td>7.2 x 10^-5</td>
<td>7.2 x 10^-5</td>
<td>7.4 x 10^-5</td>
<td>1.71 x 10^-6</td>
</tr>
<tr>
<td><em><em>Elect. Conduct. (% IACS</em>)</em>*</td>
<td></td>
<td></td>
<td></td>
<td>101</td>
</tr>
<tr>
<td><strong>Therm. Conduct. (W/m-K)</strong></td>
<td>16.2</td>
<td>16.3</td>
<td>16.0</td>
<td>391</td>
</tr>
<tr>
<td><strong>Coeff. Of Therm. Exp. (°C^-1)</strong></td>
<td>17.2x10^-6</td>
<td>16.0x10^-6</td>
<td>16.0x10^-6</td>
<td>17.5x10^-6</td>
</tr>
</tbody>
</table>
Tubing - Seamless and Welded

Seamless (extruded)

Cleaner to start with and easier to clean

Welded (rolled & welded)

Rolling can embed dirt in the surface

January 19-23 2015
Stainless steels are the most common material for making knife-edge sealing flanges.

For making knife-edge seal flanges, either ESR or cross-forged stainless steels should be used to avoid costly defects on the knife-edge tip.

The electroslag remelting (ESR) process is used to remelt and refine steels and various super-alloys, resulting in high-quality ingots.
Cornell DC Photo-Cathode Electron Gun Chamber

As received stainless steel gun chamber

Air-baked, and assembled to the gun, w/ pumps and gauges

January 19-23 2015
Aluminum Alloys

- Aluminum alloys are widely used in electron/positron storage rings, due to their high electric and thermal conductivity.

- Another key feature of (some) aluminum alloys is their extrudability, to form complex beam pipe shapes for cooling, ante-chamber, and vacuum distributed pumping.

- Most aluminum alloys are weldable, thought more difficult then stainless steel. Brazability is very poor for aluminum alloys.

- Most aluminum alloys will not be magnetized from cold-world (bending, etc.) and/or welding.

- Aluminum alloys are usually not form long lifetime radioactivity.

- However, the relatively low strength and hardness prevent aluminum alloys to be widely used for all-metal sealing flanges.
### Aluminum and Alloys

<table>
<thead>
<tr>
<th>Alloy Number</th>
<th>Major Alloy Element(s)</th>
<th>Characteristics and Sample Applications</th>
</tr>
</thead>
<tbody>
<tr>
<td>1xxx</td>
<td>None</td>
<td>Good electric and thermal conductivities, corrosion resistance. Typical applications: electric conductor wires and bus</td>
</tr>
<tr>
<td>2xxx</td>
<td>Copper</td>
<td>High strength, at room and elevated temperatures, Alloys 2011, 2017, and 2117 are widely used for fasteners and screw-machine stock</td>
</tr>
<tr>
<td>3xxx</td>
<td>Manganese</td>
<td>Similar property as 1100, slightly higher strength. Good for sheet works.</td>
</tr>
<tr>
<td>4xxx</td>
<td>Silicon</td>
<td>Excellent flow characteristics. Alloy 4032 for forging, 4043 used for GMAW and TIG 6xxx alloys.</td>
</tr>
<tr>
<td>5xxx</td>
<td>Magnesium</td>
<td>Mostly for structural applications, matching 6xxx extrusions well. 5083 alloy suitable for cryogenic applications.</td>
</tr>
<tr>
<td>6xxx</td>
<td>Magnesium + Silicon</td>
<td>6061-T6 is one of the most commonly used, 6063 is mostly used in extruded shapes</td>
</tr>
<tr>
<td>7xxx</td>
<td>Zinc</td>
<td>Mostly for structural, supporting frames. 7075-T6 is one of the most used 7000 series aluminum, and strongest one.</td>
</tr>
<tr>
<td>8xxx</td>
<td>Sn, Li, etc.</td>
<td>Specialty alloys not cover by other series.</td>
</tr>
</tbody>
</table>
Aluminum and Alloys – Tempers

- **Besides alloy designations (with Nxxx), there are standard temper designations for aluminum alloys, with one letter and a numeral.**

- **For strain-hardened (cold-worked), there are hardness designations.**
  - **-F** as fabricated;
  - **-H1** Strain hardened without thermal treatment
  - **-H2** Strain hardened and partially annealed
  - **-H3** Strain hardened and stabilized by low temperature heating

- **For alloys can be heat treated to produce stable tempers (partial list)**
  - **-O** Full soft (annealed);
  - **-T2** Cooled from hot working, cold-worked, and naturally aged
  - **-T4** Solution heat treated and naturally aged
  - **-T5** Cooled from hot working and artificially aged (at elevated temperature)
    - **-T51** Stress relieved by stretching
    - **-T511** Minor straightening after stretching
  - **-T6** Solution heat treated and artificially aged
    - **-T651** Stress relieved by stretching
Aluminum and Alloys - General Characteristics

- Moderate strength, good formability, easy to machine
- 6063-T4 can be extruded in complicated shapes
- 6061-T6 is the most common aluminum alloy for vacuum components
- 5083 is a good alloy for welding
- Aluminum is much cheaper to machine than stainless steel (2x to 3x cheaper)
- Aluminum is much less likely been radiactivated.
- Special care must be taken in the design of welds and the techniques used due to higher thermal conductivity and thermal expansion (30% > SS)
- Surface anodizing degrades outgassing characteristics, but improves chemical resistance
# Aluminum and Alloys – A Quick Comparison

<table>
<thead>
<tr>
<th>Alloy</th>
<th>Formability Workability</th>
<th>Weldability</th>
<th>Machinability</th>
<th>Heat Treatable</th>
<th>Strength</th>
<th>Corrosion Resistance</th>
</tr>
</thead>
<tbody>
<tr>
<td>1100</td>
<td>Excellent</td>
<td>Excellent</td>
<td>Good</td>
<td>No</td>
<td>Low</td>
<td>Excellent</td>
</tr>
<tr>
<td>2011</td>
<td>Good</td>
<td>Poor</td>
<td>Excellent</td>
<td>Yes</td>
<td>High</td>
<td>Poor</td>
</tr>
<tr>
<td>2024</td>
<td>Good</td>
<td>Poor</td>
<td>Fair</td>
<td>Yes</td>
<td>High</td>
<td>Poor</td>
</tr>
<tr>
<td>3003</td>
<td>Excellent</td>
<td>Excellent</td>
<td>Good</td>
<td>No</td>
<td>Medium</td>
<td>Good</td>
</tr>
<tr>
<td>5052</td>
<td>Good</td>
<td>Good</td>
<td>Fair</td>
<td>No</td>
<td>Medium</td>
<td>Excellent</td>
</tr>
<tr>
<td>6061</td>
<td>Good</td>
<td>Good</td>
<td>Good</td>
<td>Yes</td>
<td>Medium</td>
<td>Excellent</td>
</tr>
<tr>
<td>6063</td>
<td>Good</td>
<td>Good</td>
<td>Good</td>
<td>Yes</td>
<td>Medium</td>
<td>Good</td>
</tr>
<tr>
<td>7075</td>
<td>Poor</td>
<td>Poor</td>
<td>Average</td>
<td>Yes</td>
<td>High</td>
<td>Fair</td>
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</tbody>
</table>
## Typical Mechanical Properties for Aluminum

<table>
<thead>
<tr>
<th>Property</th>
<th>1100-0</th>
<th>5083-H34</th>
<th>6061-T6</th>
<th>OFE Cu</th>
</tr>
</thead>
<tbody>
<tr>
<td>Tensile Strength (MPa)</td>
<td>165</td>
<td>345</td>
<td>310</td>
<td>338</td>
</tr>
<tr>
<td>Tensile Strength (ksi)</td>
<td>23.9</td>
<td>50.0</td>
<td>45.0</td>
<td>49.0</td>
</tr>
<tr>
<td>Yield Strength (Mpa)</td>
<td>150</td>
<td>280</td>
<td>275</td>
<td>217</td>
</tr>
<tr>
<td>Yield Strength (ksi)</td>
<td>21.8</td>
<td>40.6</td>
<td>39.9</td>
<td>31.5</td>
</tr>
<tr>
<td>Elongation (%)</td>
<td>5</td>
<td>9</td>
<td>12</td>
<td>55</td>
</tr>
<tr>
<td>Modulus of Elasticity (Mpa)</td>
<td>69</td>
<td>70.3</td>
<td>69</td>
<td>115</td>
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<tr>
<td>Modulus of Elasticity (ksi)</td>
<td>10.0</td>
<td>10.2</td>
<td>10.0</td>
<td>16.7</td>
</tr>
</tbody>
</table>

Ref. www.matls.com
# Typical Physical Properties for Aluminum

<table>
<thead>
<tr>
<th>Property</th>
<th>1100-0</th>
<th>5083-H34</th>
<th>6061-T6</th>
<th>OFE Cu</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Composition:</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Al 99%</td>
<td>Al 94.8%</td>
<td>Cu 0.1%</td>
<td>Al 98%</td>
<td>Cu 100%</td>
</tr>
<tr>
<td>Cu 0.05-0.2%</td>
<td>Cr 0.05-0.25%</td>
<td>Mn 0.4-1%</td>
<td>Cu 0.15-0.4%</td>
<td></td>
</tr>
<tr>
<td>Mn 0.05%</td>
<td>Mg 4-4.9%</td>
<td>Cr 0.04-0.35%</td>
<td>Cr 0.04-0.35%</td>
<td></td>
</tr>
<tr>
<td>Si+Fe 0.95%</td>
<td>Mn 0.4-1%</td>
<td>Mg 0.8-1.2%</td>
<td>Mg 0.8-1.2%</td>
<td></td>
</tr>
<tr>
<td>Zn 0.1%</td>
<td>Fe 0.4%</td>
<td>Mn 0.15%</td>
<td>Mn 0.15%</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Si 0.4%</td>
<td>Fe 0.7%</td>
<td>Si 0.4-0.8%</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Ti 0.15%</td>
<td>Ti 0.15%</td>
<td>Ti 0.15%</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Zn 0.25%</td>
<td>Zn 0.25%</td>
<td>Zn 0.25%</td>
<td></td>
</tr>
<tr>
<td><strong>Melting Point (°C)</strong></td>
<td>643</td>
<td>591</td>
<td>582</td>
<td>1083</td>
</tr>
<tr>
<td><strong>Density (g/cc)</strong></td>
<td>2.71</td>
<td>2.66</td>
<td>2.7</td>
<td>8.92</td>
</tr>
<tr>
<td><strong>Electrical Resistivity (Ω-cm)</strong></td>
<td>3x10⁻⁶</td>
<td>5.9x10⁻⁶</td>
<td>3x10⁻⁶</td>
<td>1.7x10⁻⁶</td>
</tr>
<tr>
<td><strong>Heat Capacity (J/g-°C)</strong></td>
<td>0.904</td>
<td>0.9</td>
<td>0.896</td>
<td>0.385</td>
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<tr>
<td><strong>Therm. Conduct. (W/m-K)</strong></td>
<td>218</td>
<td>117</td>
<td>167</td>
<td>391</td>
</tr>
<tr>
<td><strong>Coeff. Of Therm. Exp. (°C⁻¹)</strong></td>
<td>25.5x10⁻⁶</td>
<td>26x10⁻⁶</td>
<td>25.2x10⁻⁶</td>
<td>17.5x10⁻⁶</td>
</tr>
</tbody>
</table>

Ref. www.matls.com
More on 6061-T6 Alloy – Heating Effects

- 6061-T6 alloy loss strength quickly at temperature above 177°C, the typical artificial aging temperature during heat-treatment.

- Heating above 150°C can anneal 6061-T6 alloy over time. So bakeout temperature should be at or below 150°C if -T6 strength is to be retained.

Ref. Properties of Aluminum Alloys, Tensile, Creep and Fatigue Data at High and Low Temperatures, by J. Kaufman, p.168
Machined Aluminum “Switch-yard” Chamber
Aluminum Electron Beam Stopper

600 kW/15 MeV Beam Stopper for Cornell Prototype ERL Injector
Extruded Beam Pipes - Complex Shape
Extruded Beam Pipes – Undulator Chamber

Base Extrusion 6061-T6
Extruded Beam Pipes – Undulator Chamber

Extrusion machined to 0.6-mm thin to allow small undulator gap

FEA and tests to ensure adequate safety factor (>2 in this case)

6061-T6 Aluminum Alloy:
Yield Tensile Strength: 276 MPa
Ultimate Tensile Strength: 310 MPa
Copper and Alloys

- Copper and alloys are designated by a system starting with letter “C”, followed by 5 digits, more copper contents with lower numbers.

- Some commonly used copper alloys are **C10100 (OFHC)**, C26800 (Yellow Brass, Zinc alloy), C61400 (Bronze, Silicon alloy), C17200 (Beryllium coppers)

- Low-to-moderate strength, good formability

- Excellent electrical and thermal characteristics

- Difficult to weld (e-beam welding is best)

- May be joined by welding, brazing, and soldering

- Good outgassing characteristics, rates can be decreased by following good machining techniques, chemical and baking (~200°C)
Oxygen-Free High Conductivity (OFHC) Copper

- **Oxygen-free high thermal conductivity (OFHC) copper generally refers to a group of wrought high conductivity copper alloys that have been electrolytically refined to reduce the level of oxygen to 0.001% or below.**

- **OFHC is often used in accelerator vacuum system, where high heat load is encountered. It is also used to construct normal conducting radio-frequency (RF) cavities.**

- **C10100** - This is the purest grade, with 99.99% Cu, <0.0005% (or 5 ppm) oxygen content.

- **C10200** - 99.95% Cu (including Ag), <0.001% (10-ppm) oxygen content.

- **C11000** - Also known as Electrolytic-Tough-Pitch (ETP) copper. It is 99.9% pure and has 0.02% to 0.04% oxygen content (typical).

- **Low oxygen content is critical for vacuum assemblies involving welding.**
### OFHC Copper Properties

<table>
<thead>
<tr>
<th>Density</th>
<th>Electric Resistivity</th>
<th>Thermal Conductivity</th>
<th>C.T.E.</th>
<th>M.P.</th>
</tr>
</thead>
<tbody>
<tr>
<td>8.9 g/cc</td>
<td>1.71x10^{-6} Ω-cm</td>
<td>383 ~ 391 W/m-K</td>
<td>17.0 µm/m-K</td>
<td>1083°C</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Temper Designation Standard</th>
<th>Tensile Strength (ksi)</th>
<th>Yield Strength (ksi, min.)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Min.</td>
<td>Max.</td>
</tr>
<tr>
<td>060 Soft</td>
<td>30</td>
<td>38</td>
</tr>
<tr>
<td>H00 Cold-Rolled, 1/8-hard</td>
<td>32</td>
<td>40</td>
</tr>
<tr>
<td>H01 Cold-rolled, ¼-hard</td>
<td>34</td>
<td>42</td>
</tr>
<tr>
<td>H02 Half Hard</td>
<td>37</td>
<td>46</td>
</tr>
<tr>
<td>H03 ¾-hard</td>
<td>41</td>
<td>50</td>
</tr>
<tr>
<td>H04 Full hard</td>
<td>43</td>
<td>52</td>
</tr>
</tbody>
</table>

- **The hard-tempers of pure coppers can only be achieved via work-hardening, i.e. cannot be hardened by heat-treatment.**
- **Annealing of pure copper starts as low as 150°C**
Copper Vacuum Chamber Example

- **C10100 $\frac{1}{2}$-Hard Cu sheet bend to U-box to form a TiSP pumping plenum**

- **C10100 $\frac{1}{2}$-Hard Cu plates machined to form a beam pipe**

- **SST L-shaped plates added to complete vacuum envelop, with enhance mechanical strength.**

  (Flanges to be added)
Copper Extrusions

- Cooling Bar Extrusion
- Screen Extrusion
- "Dipole" Chamber Extrusion
Machined Copper Chamber (PEP-II RF Cavities)

- 26 cavities
- $4M total fabrication cost
- Integral cooling channels with electroformed cover
- 5 axis machining
- e-beam welded
- 17 separate manufacturing steps
Copper Coils for a Positron Converter @ CESR

- Pulsed current flowing through a two-layer coil to focus positrons from the target to down-stream accelerator.

- Tubular copper conductor in full annealed temper for winding.

- The formed coil (with vacuum brazed end caps) must be hardened to withhold turn-to-turn pulsing forces.

- The hardening was achieved by manual stretch-compression cycles.
Copper Strengthening

- OFHC coppers are commonly used for construction of accelerator vacuum chambers for their excellent thermal and electric conductivity. Comparing to aluminum alloys, OFHC coppers also provide better radiation (especially gamma) shielding.

- However, pure copper has relatively low strength, even hardened. For applications with higher stresses (especially cyclic loading), the strength of copper can be improved by various strengthening mechanisms.

- Precipitation hardening (PH) coppers are heat-treatable to very much higher strength, while retaining its high electric and thermal conductivities. A commonly used PH copper is CuCrZr (UNS C18150).

- Dispersion-strengthened (DS) coppers are among the other commercially available materials, GlidCop®, Al15, Al25 and Al60.

Mechanical Strength vs. Thermal Conductivities

Mechanical Strength at High Temperatures

Fracture Toughness

- Black = CuAl15 or CuAl25
- Red = CuCrZr
- Green = CuNiBe

\( J_a \) (kJ/m\(^2\)) vs Temperature (°C)
Cornell Undulator Chamber made of CuCrZr

PH Copper CuCrZr has excellent mechanical and thermal properties, thus was chosen for this thin-wall chamber. However, both Cr and Zr alloying elements may become radioactive by lost particles.
Glidcop is pure copper with Al₂O₃ dispersed throughout.

- High strength, moderate formability, poor weldability.
- Available in sheets, plate, wire, and extruded rounds.
- Maintains good mechanical strength after brazing.
- Outgassing rates are similar to pure copper.
- Thermal and electrical properties are good.

<table>
<thead>
<tr>
<th>Grade Designations</th>
<th>Copper</th>
<th>Al₂O₃</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Wt %</td>
<td>Vol %</td>
</tr>
<tr>
<td>UNS C15715 Glidcop AL-15</td>
<td>99.7</td>
<td>99.3</td>
</tr>
<tr>
<td>UNS C15725 Glidcop AL-25</td>
<td>99.5</td>
<td>98.8</td>
</tr>
<tr>
<td>UNS C15760 Glidcop AL-60</td>
<td>98.9</td>
<td>97.3</td>
</tr>
</tbody>
</table>

Ref. SCM Metal Products
## Glidcop™ Physical Properties

<table>
<thead>
<tr>
<th>Property</th>
<th>C15715</th>
<th>C15725</th>
<th>C15760</th>
<th>OFE Cu</th>
</tr>
</thead>
<tbody>
<tr>
<td>Melting Point (°C)</td>
<td>1083</td>
<td>1083</td>
<td>1083</td>
<td>1083</td>
</tr>
<tr>
<td>Density (lb/in³)</td>
<td>0.321</td>
<td>0.320</td>
<td>0.318</td>
<td>0.323</td>
</tr>
<tr>
<td>Electrical Resistivity (Ω)</td>
<td>11.19</td>
<td>11.91</td>
<td>13.29</td>
<td>10.20</td>
</tr>
<tr>
<td>Elect. Conduct. (% IACS*)</td>
<td>92</td>
<td>87</td>
<td>78</td>
<td>101</td>
</tr>
<tr>
<td>Therm. Conduct. (W/m-K)</td>
<td>365</td>
<td>344</td>
<td>322</td>
<td>391</td>
</tr>
<tr>
<td>Coeff. Of Therm. Exp. (°C⁻¹)</td>
<td>16.6x10⁻⁶</td>
<td>16.6x10⁻⁶</td>
<td>16.6x10⁻⁶</td>
<td>17.7x10⁻⁶</td>
</tr>
<tr>
<td>Mod. Of Elasticity (psi)</td>
<td>19x10⁶</td>
<td>19x10⁶</td>
<td>19x10⁶</td>
<td>19x10⁶</td>
</tr>
</tbody>
</table>

* International Annealed copper Standard

Ref. SCM Metal Products
## Glidcop™ Mechanical Properties

<table>
<thead>
<tr>
<th>Grade</th>
<th>Form</th>
<th>Tensile Strength (ksi)</th>
<th>Yield Strength (ksi)</th>
</tr>
</thead>
<tbody>
<tr>
<td>AL-15 (C15715)</td>
<td>Plate</td>
<td>53 ~ 70</td>
<td>37 ~ 66</td>
</tr>
<tr>
<td></td>
<td>Rod</td>
<td>57 ~ 72</td>
<td>47 ~ 66</td>
</tr>
<tr>
<td></td>
<td>Rounds</td>
<td>53</td>
<td>37</td>
</tr>
<tr>
<td>AL-25 (C15725)</td>
<td>Plate</td>
<td>60 ~ 76</td>
<td>43 ~ 68</td>
</tr>
<tr>
<td></td>
<td>Rod</td>
<td>64 ~ 80</td>
<td>52 ~ 77</td>
</tr>
<tr>
<td></td>
<td>Rounds</td>
<td>60</td>
<td>43</td>
</tr>
<tr>
<td>AL-60 (C15760)</td>
<td>Plate</td>
<td>--</td>
<td>--</td>
</tr>
<tr>
<td></td>
<td>Rod</td>
<td>72 ~ 90</td>
<td>69 ~ 87</td>
</tr>
<tr>
<td></td>
<td>Rounds</td>
<td>68</td>
<td>48</td>
</tr>
<tr>
<td>C10100</td>
<td></td>
<td>30 ~ 50</td>
<td>20 ~ 35</td>
</tr>
</tbody>
</table>

1. Ref. [http://www.hoganas.com](http://www.hoganas.com)
2. Large spread reflect strength at different tempers
NSLS II Crotch and Absorber Made of Glidcop™

Insertable Crotch

Stick Photon Absorber

Ref: H. Hseuh, NSLS II, BNL
Other Metals – Beryllium

- Beryllium is the lightest metal with good mechanical strength and good thermal conductivity.
- Beryllium is machinable and can be jointed via vacuum braze or e-beam welding.
- Beryllium is hazard mat’l, must be handled by highly trained experts.

UHV X-ray Windows

75µm Be e⁻ Injection Window

SynchLight Mirror

Be Beampipe for CMS
Niobium & Titanium

- High purity, small grain niobium is the key material for constructing superconducting RF cavities for many existing and future (such as Jlab, Cornell CESR and ERL, Tesla, ILC, etc.) facilities.

- The grade of the Nb material is usually certified by so-called RRR (Residual-resistance ratio). Hydrogen in the Nb bulk is often degassed for high-Q cavities.

- To match CTE, titanium (grade 1 and grade 5) are used to joint to Nb cavities for flanges and helium vessels.

- E-beam welding is the primary technique for Nb and Ti.
C.T.E. of some metals

![Graph showing the linear thermal expansion coefficient of various metals as a function of temperature.](image)
Non-Metals – Ceramics and Glasses

- Alumina ceramics (Al$_2$O$_3$ > 99%) are widely used for electric breaks, instrument and electric power feedthroughs, RF windows in the accelerator vacuum systems.

- Alumina ceramic beam pipes with thin inner metallic coating are also used as a part of pulsed magnet for beam feedback, injection kickers, etc.

- Ceramics are jointed to metal flanges using vacuum furnace braze technique.

- Many type of glasses are used mainly as viewports on vacuum systems, for visual inspection of in-vacuum components, for light transmissions (laser entrance, beam profile viewers, etc.)

- Machined ceramic parts are UHV compatible. Special diamond-tipped tools are used for ceramic machining. There are also machinable ceramics, such Macor®.
## Properties of Some Glasses for Vacuum

<table>
<thead>
<tr>
<th>Property</th>
<th>Fused Silica</th>
<th>Pyrex 7740</th>
<th>Pyrex 7720&lt;sup&gt;a&lt;/sup&gt;</th>
<th>Soda 7052&lt;sup&gt;a&lt;/sup&gt;</th>
<th>0080</th>
<th>0120</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Composition</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>SiO&lt;sub&gt;2&lt;/sub&gt;</td>
<td>100</td>
<td>81</td>
<td>73</td>
<td>65</td>
<td>73</td>
<td>56</td>
</tr>
<tr>
<td>B&lt;sub&gt;2&lt;/sub&gt;O&lt;sub&gt;3&lt;/sub&gt;</td>
<td>13</td>
<td>15</td>
<td>18</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Na&lt;sub&gt;2&lt;/sub&gt;O</td>
<td>4</td>
<td>4</td>
<td>2</td>
<td>17</td>
<td>4</td>
<td>4</td>
</tr>
<tr>
<td>Al&lt;sub&gt;2&lt;/sub&gt;O&lt;sub&gt;3&lt;/sub&gt;</td>
<td>2</td>
<td>2</td>
<td>7</td>
<td>1</td>
<td>2</td>
<td>2</td>
</tr>
<tr>
<td>K&lt;sub&gt;2&lt;/sub&gt;O</td>
<td></td>
<td></td>
<td>3</td>
<td></td>
<td></td>
<td>9</td>
</tr>
<tr>
<td>PbO</td>
<td></td>
<td></td>
<td>6</td>
<td></td>
<td></td>
<td>29</td>
</tr>
<tr>
<td>LiO</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>1</td>
<td></td>
</tr>
<tr>
<td>Other</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>3</td>
<td>9</td>
</tr>
<tr>
<td><strong>Viscosity characteristics</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Strain point °C</td>
<td>956</td>
<td>510</td>
<td>484</td>
<td>436</td>
<td>473</td>
<td>395</td>
</tr>
<tr>
<td>Annealing point °C</td>
<td>1084</td>
<td>560</td>
<td>523</td>
<td>480</td>
<td>514</td>
<td>435</td>
</tr>
<tr>
<td>Softening point °C</td>
<td>1580</td>
<td>821</td>
<td>755</td>
<td>712</td>
<td>696</td>
<td>630</td>
</tr>
<tr>
<td>Working point °C</td>
<td>---</td>
<td>1252</td>
<td>1146</td>
<td>1128</td>
<td>1005</td>
<td>985</td>
</tr>
<tr>
<td>Expansion coefficient ×10&lt;sup&gt;-7&lt;/sup&gt;/°C</td>
<td>3.5</td>
<td>35</td>
<td>43</td>
<td>53</td>
<td>105</td>
<td>97</td>
</tr>
<tr>
<td>Shock temperature, 1/4-in. plate °C</td>
<td>1000</td>
<td>130</td>
<td>130</td>
<td>100</td>
<td>50</td>
<td>50</td>
</tr>
<tr>
<td>Specific gravity</td>
<td>2.20</td>
<td>2.23</td>
<td>2.35</td>
<td>2.27</td>
<td>2.47</td>
<td>3.05</td>
</tr>
</tbody>
</table>

*Source.* Reprinted with permission from Corning Glass Works, Corning, NY.

<sup>a</sup> 7720 glass is used for sealing to tungsten and 7052 glass is used for sealing to Kovar.
### Table 16.6 Physical Properties of Some Ceramics

<table>
<thead>
<tr>
<th>Ceramic</th>
<th>Main Body Composition</th>
<th>Expansion Coefficient ($\times 10^{-\text{7}}$)</th>
<th>Softening Temperature (°C)</th>
<th>Tensile Strength ($10^6$ kg/m²)</th>
<th>Specific Gravity</th>
</tr>
</thead>
<tbody>
<tr>
<td>Steatite</td>
<td>MgOSiO₂</td>
<td>70–90</td>
<td>1400</td>
<td>6</td>
<td>2.6</td>
</tr>
<tr>
<td>Forsterite</td>
<td>2MgOSiO₂</td>
<td>90–120</td>
<td>1400</td>
<td>7</td>
<td>2.9</td>
</tr>
<tr>
<td>Zircon porcelain</td>
<td>ZnO₂SiO₂</td>
<td>30–50</td>
<td>1500</td>
<td>8</td>
<td>3.7</td>
</tr>
<tr>
<td>85% alumina</td>
<td>Al₂O₃</td>
<td>50–70</td>
<td>1400</td>
<td>14</td>
<td>3.4</td>
</tr>
<tr>
<td>95% alumina</td>
<td>Al₂O₃</td>
<td>50–70</td>
<td>1650</td>
<td>18</td>
<td>3.6</td>
</tr>
<tr>
<td>98% alumina</td>
<td>Al₃O₅</td>
<td>50–70</td>
<td>1700</td>
<td>20</td>
<td>3.8</td>
</tr>
<tr>
<td>Pyroceram 9696&lt;sup&gt;a&lt;/sup&gt;</td>
<td>Corderite ceramic</td>
<td>57</td>
<td>1250</td>
<td>14&lt;sup&gt;b&lt;/sup&gt;</td>
<td>2.6</td>
</tr>
<tr>
<td>Macor 9658&lt;sup&gt;a&lt;/sup&gt;</td>
<td>Flurophlogophite</td>
<td>94</td>
<td>800</td>
<td>10&lt;sup&gt;b&lt;/sup&gt;</td>
<td>2.52</td>
</tr>
</tbody>
</table>


<sup>a</sup> Reprinted with permission from Corning Glass Works, Corning, NY.

<sup>b</sup> Modulus of rupture.
Non-Metals – Elastomers and Polymers

- Elastomers, polymers and plastics have also found application in accelerator vacuum systems. Their vacuum properties and radiation resistance must be verified for the applications.

- Elastomers, particularly Viton® (fluorocarbon) are usually used as vacuum seals, often as gate seals for UHV gate valves.

- Though Teflon is UHV compatible, it is easily hardened and break down under radiation. PEEK (Polyether ether ketone) is a type of engineered plastics that is suitable for accelerator UHV applications. PEEK has good formability and machinability. The most uses are in vacuum multi-pin connectors.

- Kapton® (polyimide) films are suitable for accelerator UHV applications.
PEEK Material Can Be ‘Gassy’

Provided by Sekiguchi Taku of CosmoTec
Dry Lubrications in UHV Systems

- For in-vacuum movement that involves two metallic surfaces in contact, particularly two similar metals, lubrication between the contacting surfaces is necessary.

- For UHV applications, dry-lubrication is widely used. In a dry lubrication, the process of lubricating relies on a solid film.

- The desirable properties of a dry lubricant are low vapor pressure, low shear strength, and good adhesion to the base metal.

- Commonly used UHV-compatible dry lubricants are silver (electroplated), MoS$_2$, WS$_2$ (Dicronite®) (via PVD). Teflon coating is also a UHV-compatible dry lubricant, however, it is not durable in radiation environment.
However, extreme low vapor pressure may not be good enough in applications where energized desorption may occur.

**UHV-compatible grease lubricants**
Stimulated desorption of Krytox

\[ m/e = 69 \ (CF_3^+) \]