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# The US Particle Accelerator School Cryosorption Pumps

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Lawrence Livermore National Laboratory

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Updated: 12/1/2014 (Yulin Li)

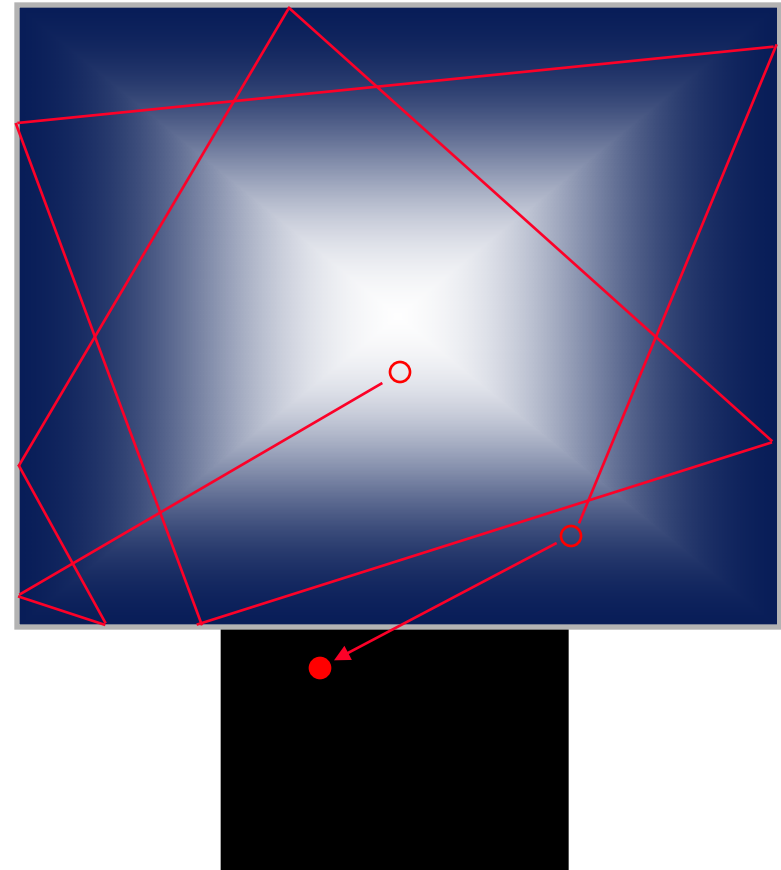


# Cryopumping Basics . . . Cryocondensation



Cooling gases to the extent that gas molecules lose sufficient energy to form condensation layers.

- A cryogenic surface will trap any molecule that contacts the surface if it is cold enough.



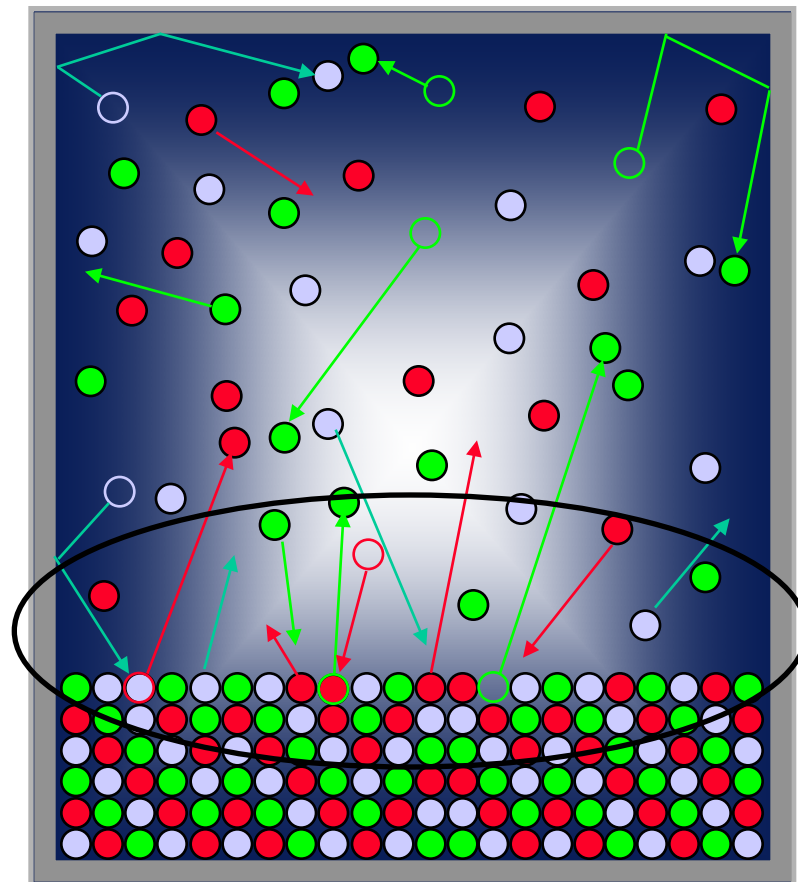
# Cryopumping Basics . . .

## Equilibrium Vapor Pressure



Equilibrium vapor pressure is the state where as many molecules are condensing as are vaporizing.

Equilibrium occurs when the rate of gas molecules returning to the liquid/solid (**condensing**) is equal to the rate of energetic molecules becoming gaseous (**vaporizing**).



# Cryopumping Basics . . .

## Pressure within a Cryopump

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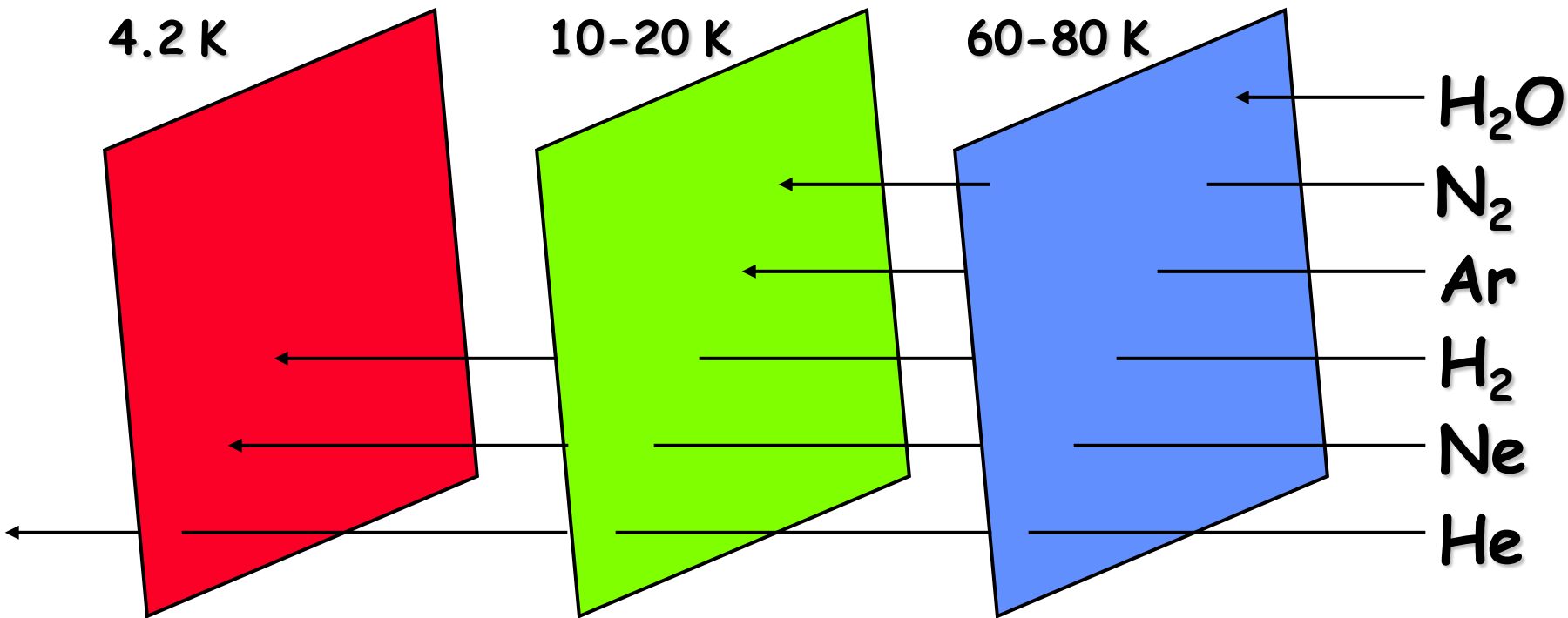


### What determines the Pressure inside a Cryopump?

Surface Temp.	at 16K	at 25K	at 31K
•Nitrogen	> $10^{-12}$ Torr	> $10^{-7}$ Torr	> $10^{-4}$ Torr
•Argon	> $10^{-12}$ Torr	> $10^{-9}$ Torr	> $10^{-4}$ Torr
•Oxygen	> $10^{-12}$ Torr	> $10^{-10}$ Torr	> $10^{-4}$ Torr
•Hydrogen	> $10^{+2}$ Torr		
•Helium	> Atm.		



# Cryopumping Basics . . . Cryocondensation

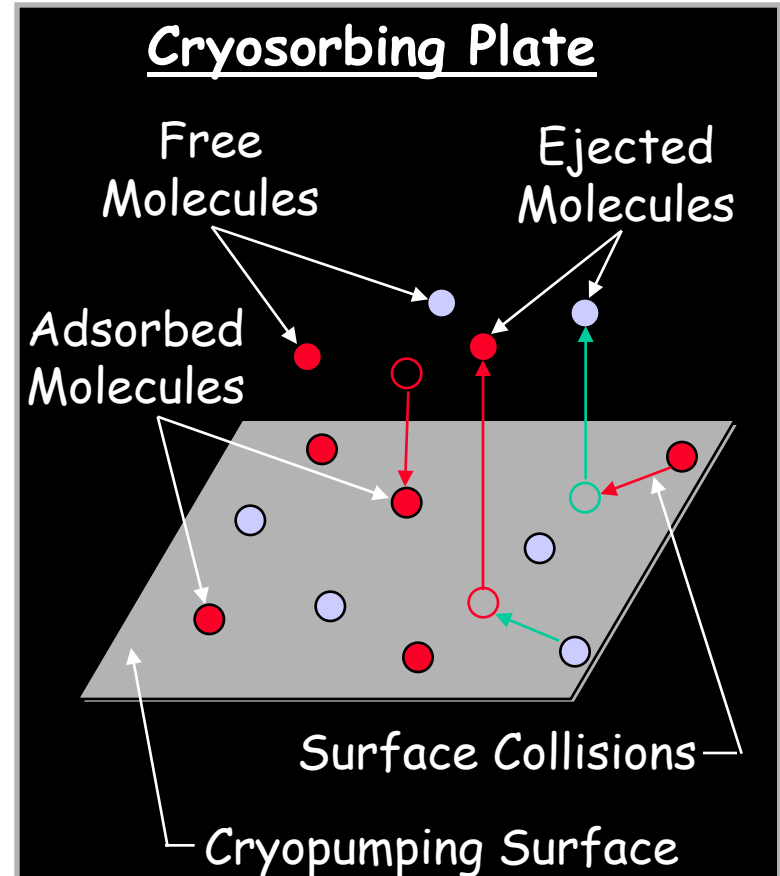




# Cryopumping Basics . . . Cryosorption

Cooling gas molecules to the extent that gas molecules, upon contacting a sufficiently cooled surface, lose enough energy to accumulate on the surface.

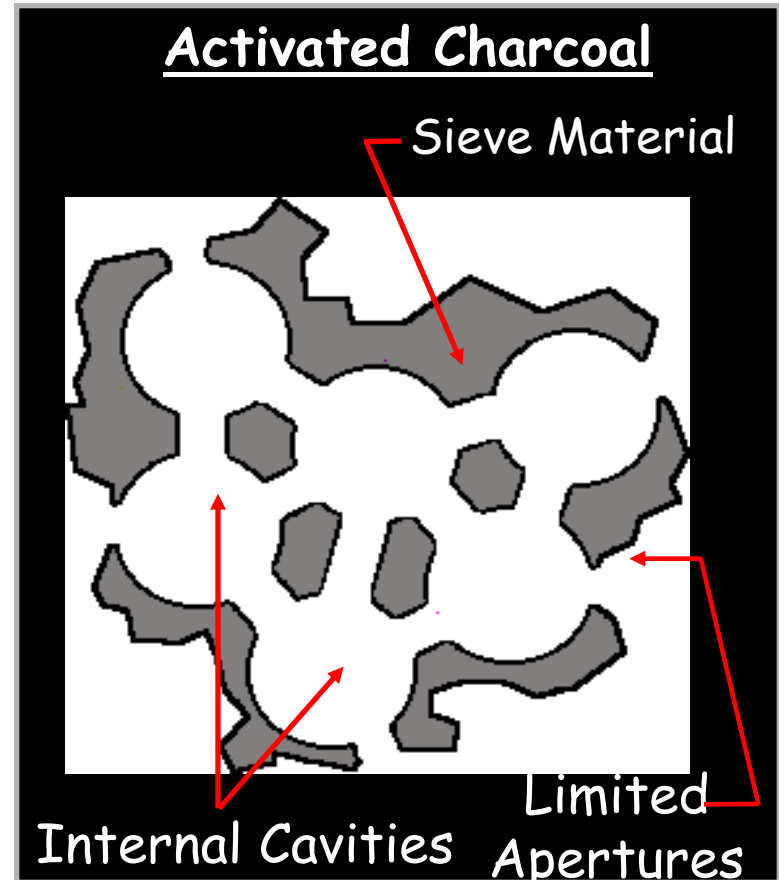
- A flat cryoadsorbing plate retains some molecules.
- Flat surface allows molecules to continue moving.



# Cryopumping Basics . . . Cryosorption



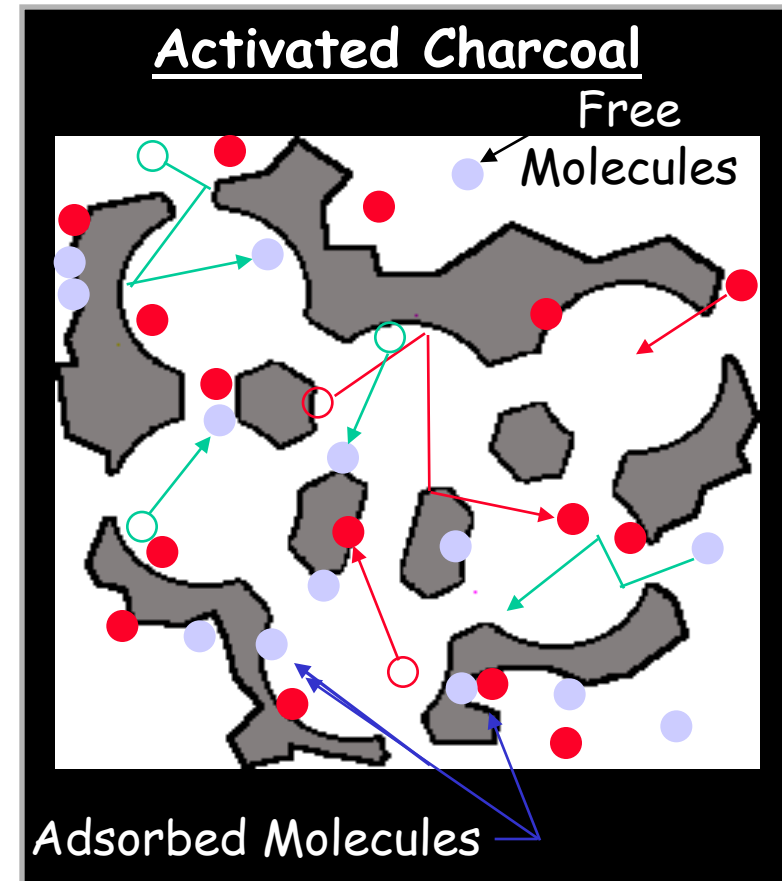
- Sieve material, such as Zeolite, charcoal, provides greater surface area and limited apertures.
- Large surface area capacity: 1150-1250 m<sup>2</sup>/gm



# Cryopumping Basics . . . Cryosorption



- Increased surface area provides greater capacity.
- Released molecules remain confined.
- Irregular surface constricts motion.
- Cryosorption of hydrogen, neon, and helium accomplished.

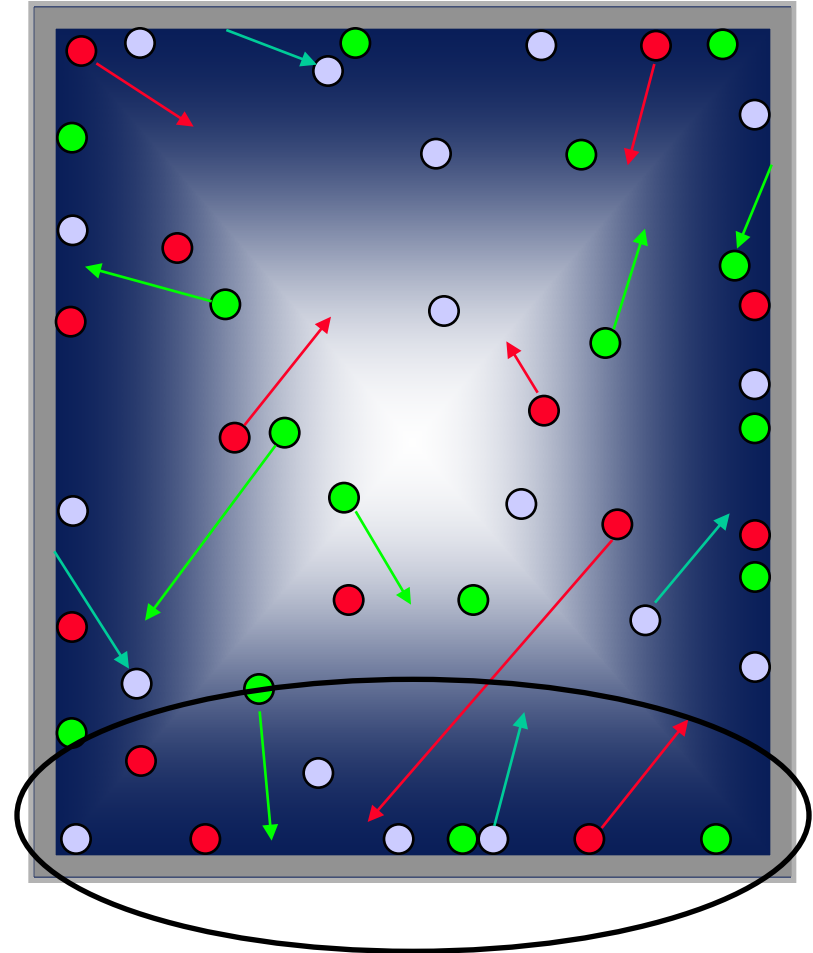




# Cryopumping Basics . . . Surface Equilibrium



When the number of molecules arriving on the chamber surface (adsorbing) equals the number leaving the surface (desorbing), then the system is in **"Surface Equilibrium"**.



# Equilibrium

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## Equilibrium Vapor Pressure:

- **CONDENSATION**
- **VAPORIZATION**

## Surface Equilibrium:

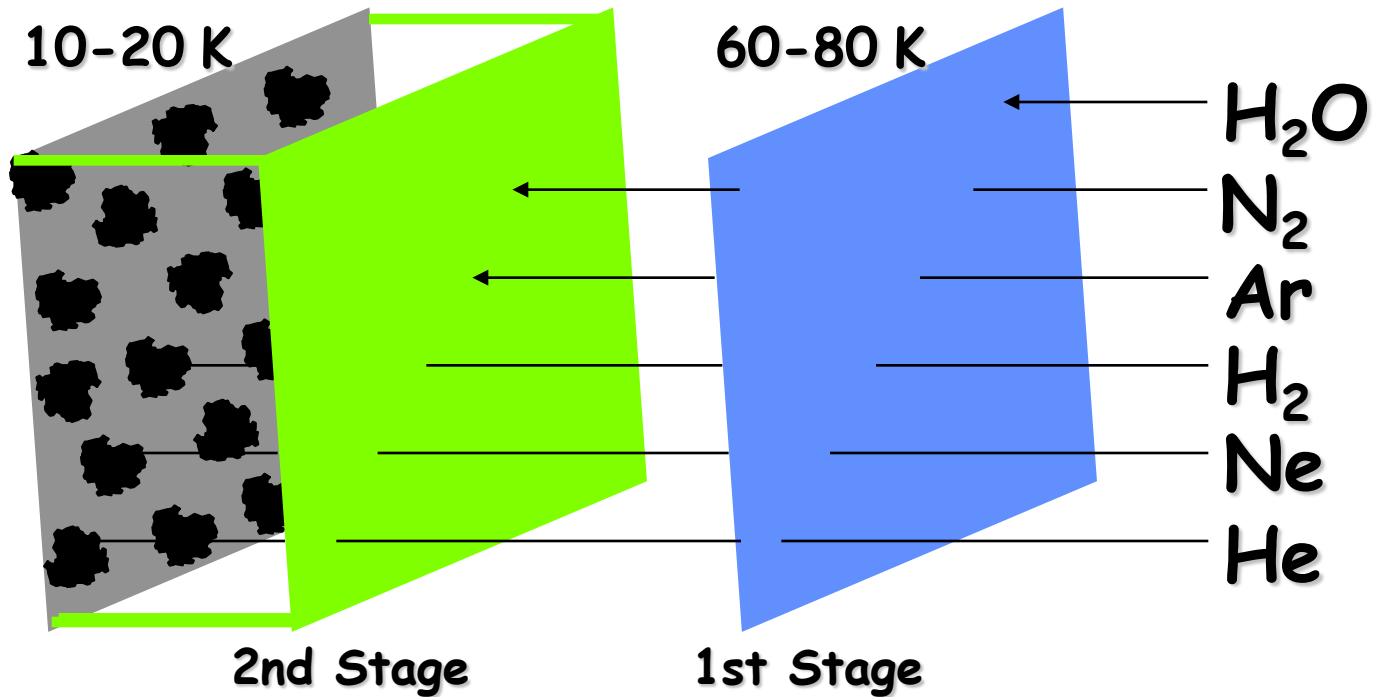
- **ADSORPTION**
- **DESORPTION**



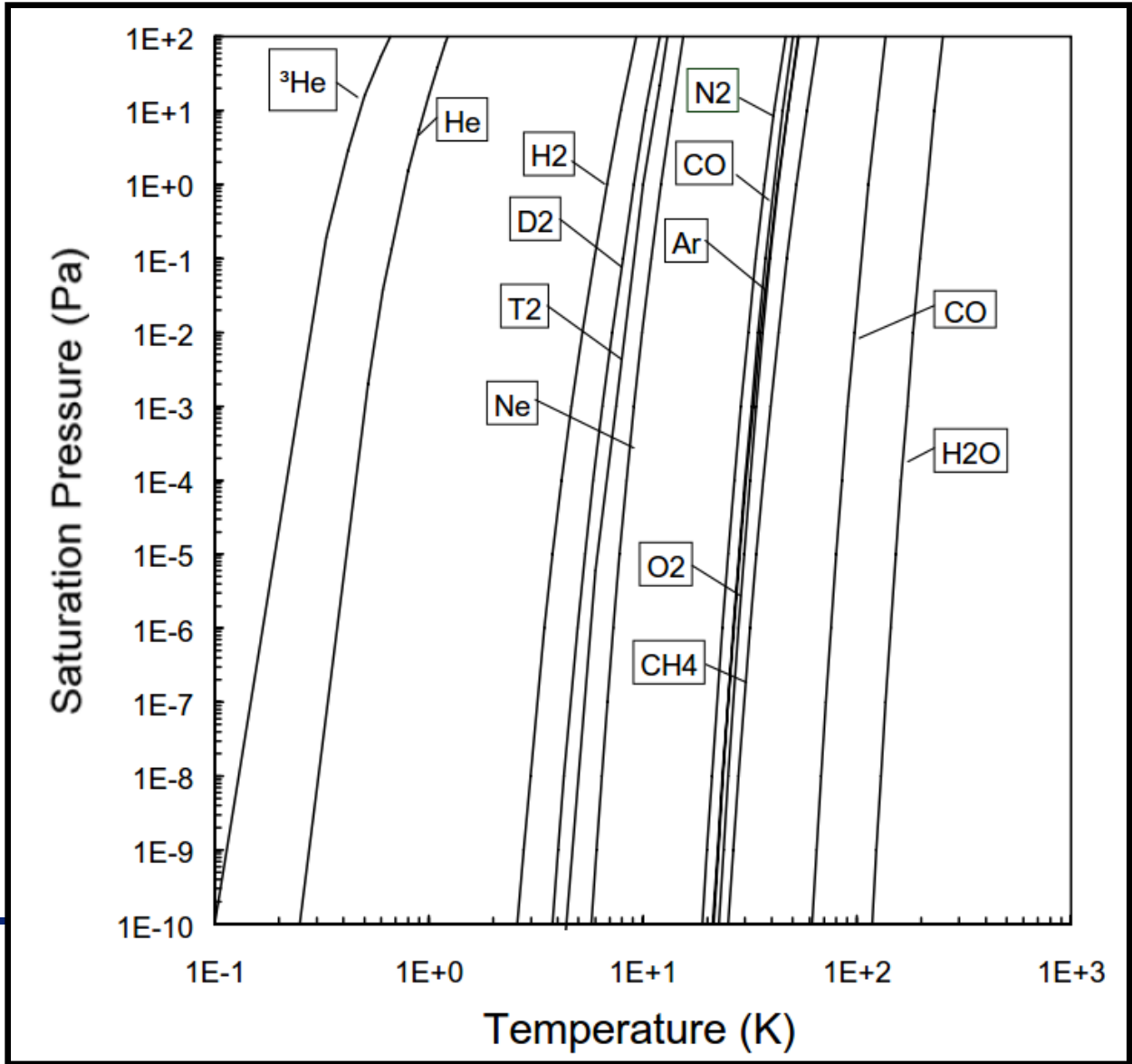
# Cryopumping Basics . . . Cryosorption and Cryocondensation



Air gases and water vapor are condensed,  
noncondensable gases are captured.



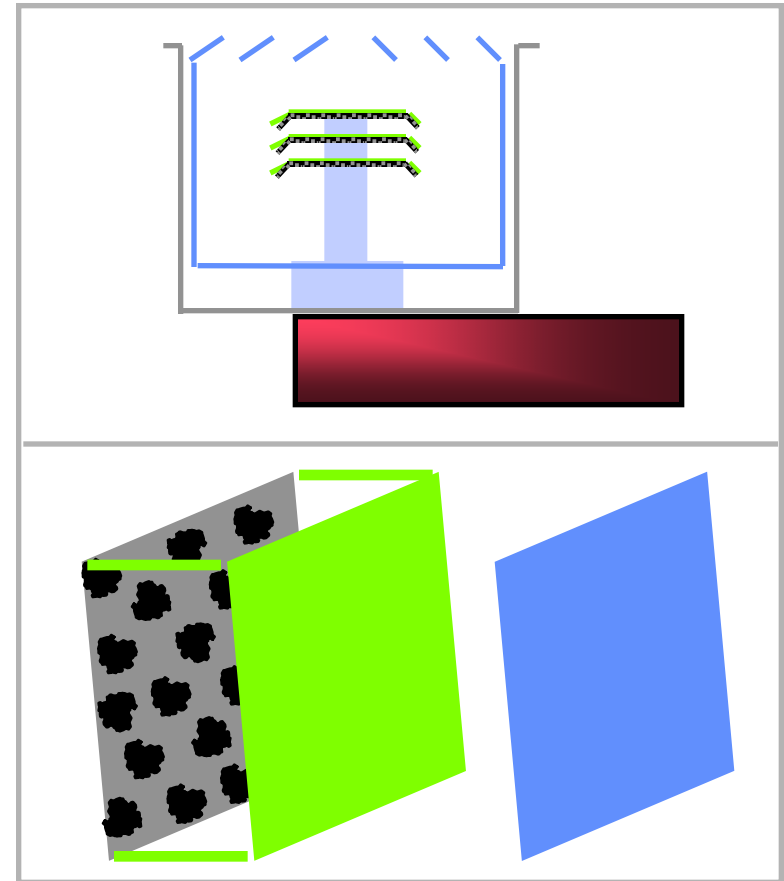
# Saturation curves of common gases



# Cryopump Concept



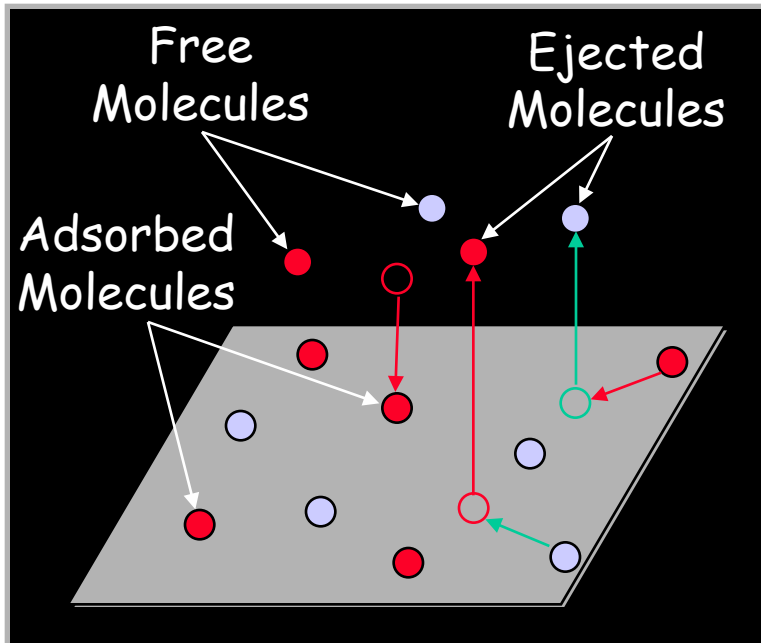
- Cryopumps are designed to create these condensing and adsorbing surfaces.



# Cryopumping Basics . . . Adsorption Isotherm



An adsorption isotherm is a measure of the surface population density of a gas at a constant temperature.



$$\sigma = f(P, T)$$

where  $\sigma$  = density of molecules of gas on a surface per  $\text{cm}^2$   
 $P$  = equilibrium pressure of system  
 $T$  = system temperature



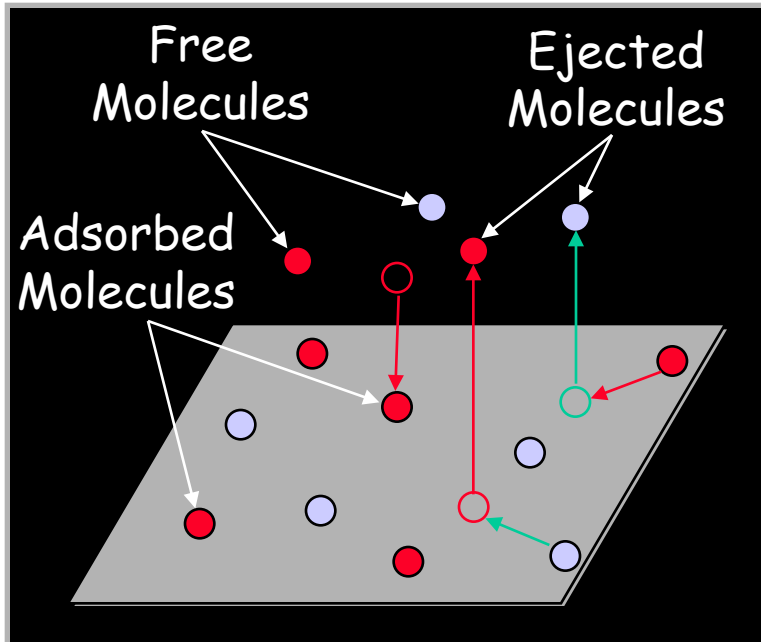
Adsorption isotherms can be expressed several ways:

## % Coverage

- $\sigma = 0.20$  surface 20% covered
- $\sigma = 1$  One monolayer ( $\sigma_m$ )
- $\sigma = 2$  Two monolayers ( $2\sigma_m$ )

## Molecules/cm<sup>2</sup>

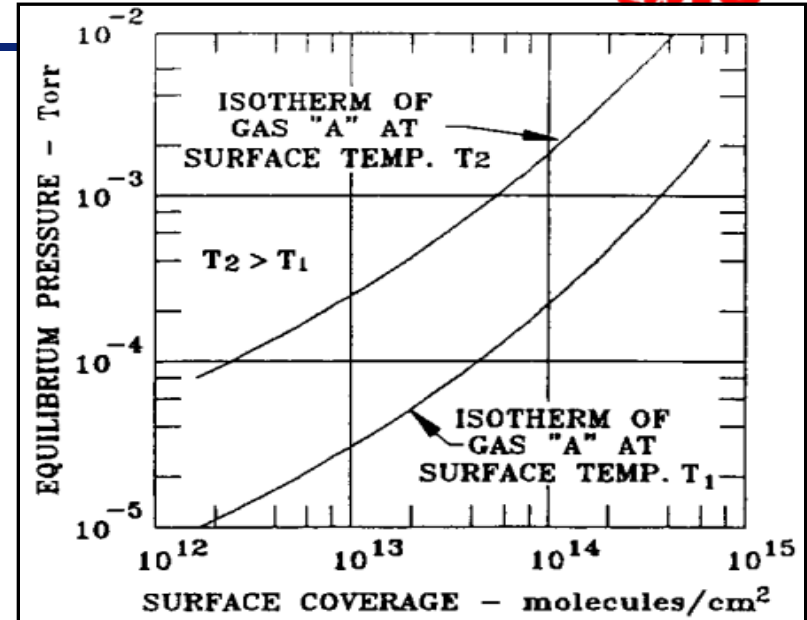
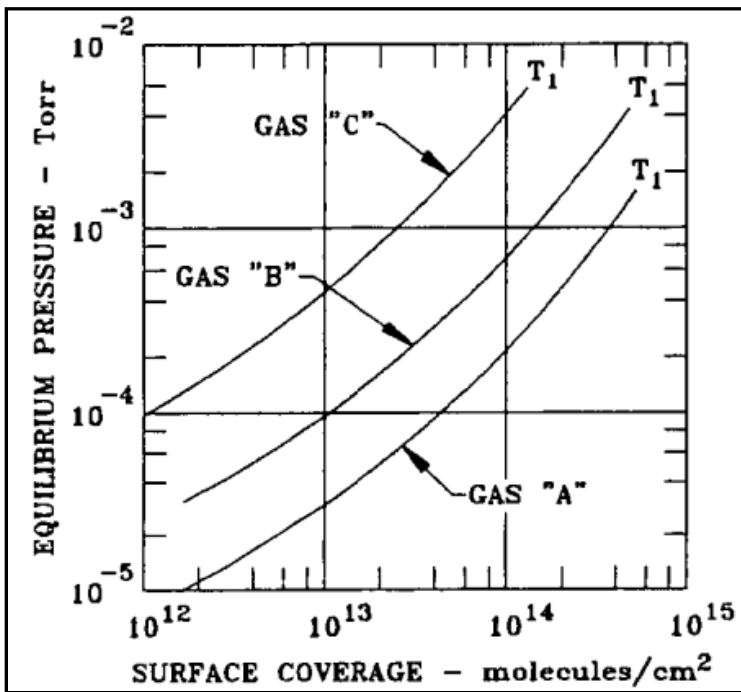
$$\sigma = 10^{15} \text{ molecules/cm}^2$$



# Cryopumping Basics . . . Adsorption Isotherm



- Usually an adsorption isotherm represents pressure vs. coverage data at a specific temperature.
- As the temperature increases, the equilibrium pressure increases for a specific surface coverage.

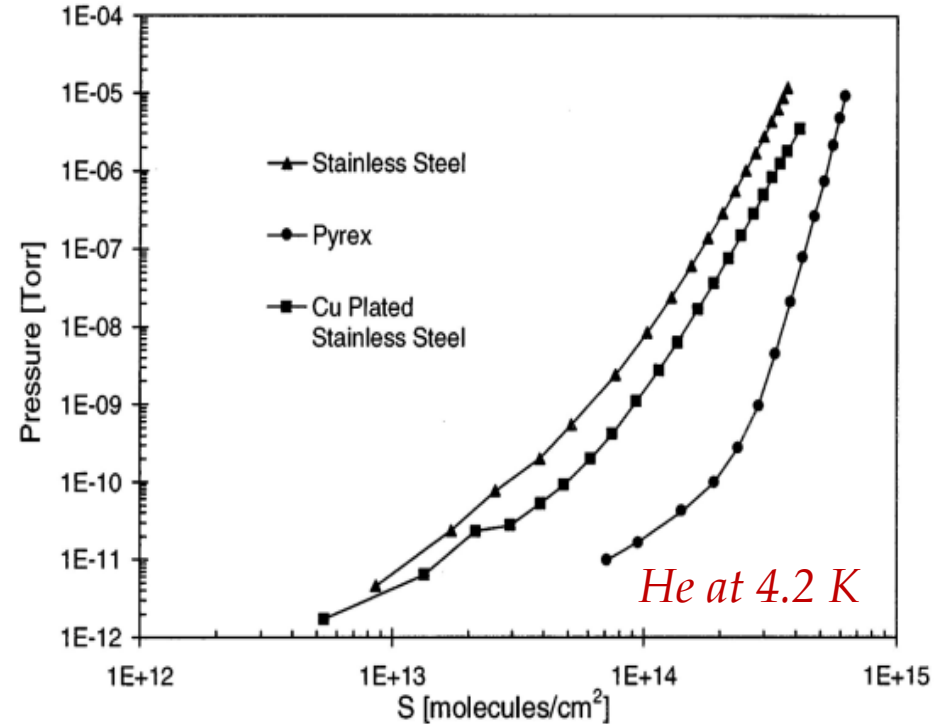
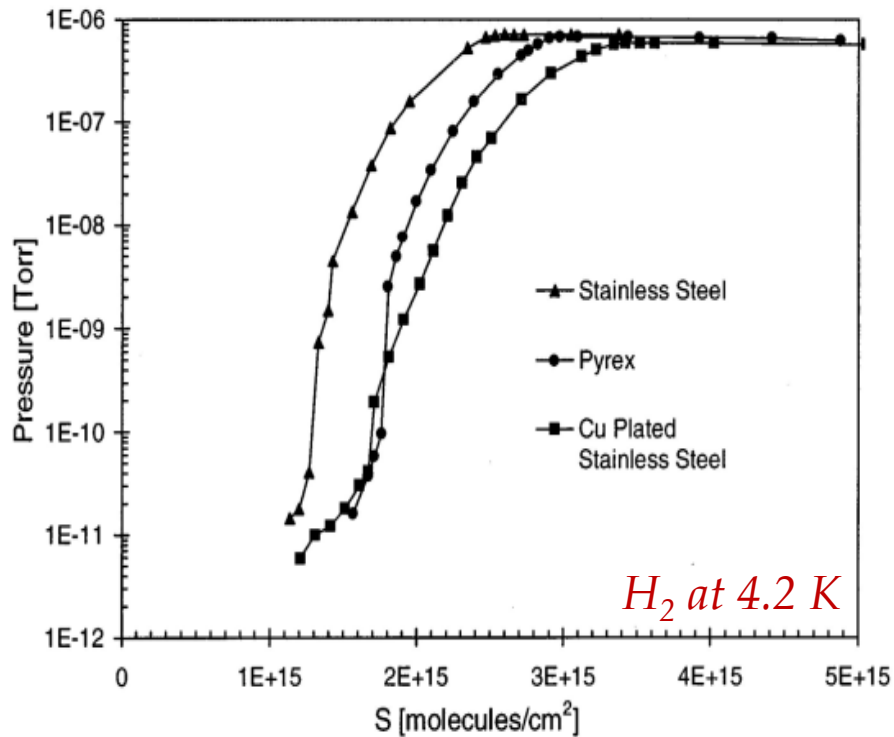


- Each gas has its own unique adsorption isotherm for the same temperature.
- For all gases, the equilibrium pressure of an adsorption isotherm is less than the vapor pressure at that temperature.
- As surface coverage goes up (to several monolayers), the equilibrium pressure will approach the vapor pressure.





# Cryopumping Basics . . . Example Isotherms



E. Wallen: "Adsorption Isotherms of He and  $H_2$  at Liquid Helium Temperature",  
JVST A15, p.265



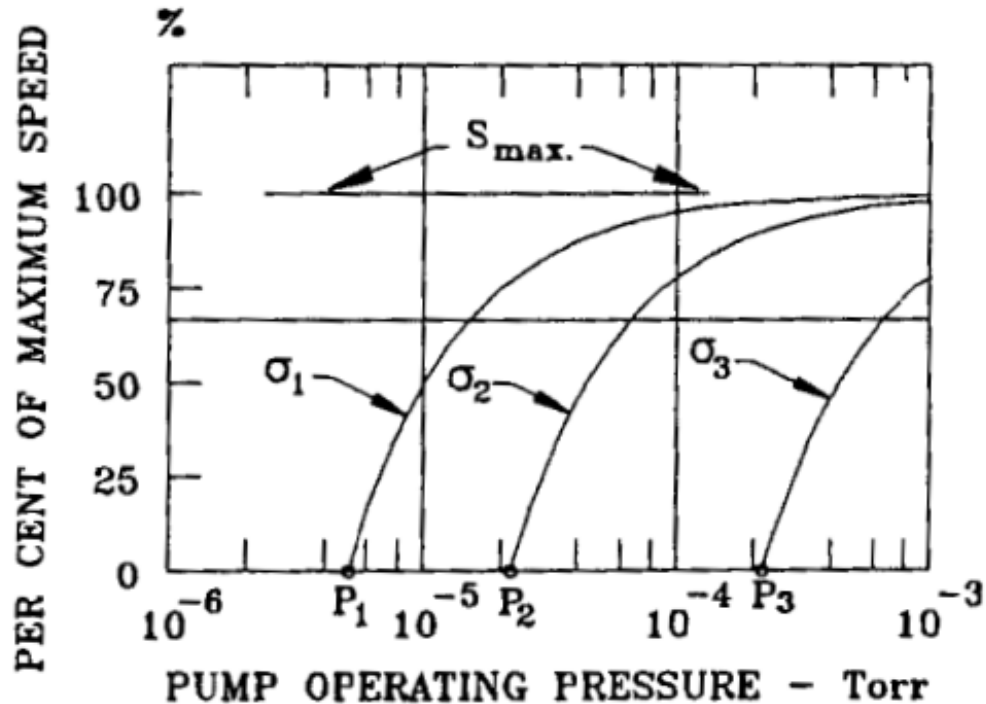
# Cryopumping Basics . . . Pumping Speed



- A cold surface has a finite pumping speed for a gas as long as the pressure of the adsorption isotherm is less than the pressure of the gas ( $P_e$ ).
- As the surface coverage increases, the equilibrium pressure increases.

$$S = S_{\max} \left( 1 - \frac{P_e}{P} \right)$$

- $S_{\max}$  is set by the surface conductance limitations of the cryopump.



In cryosorption pumping, speed is dependent on the quantity of gas already adsorbed and the pressure. That is, a cryopump has a finite capacity.

$$\sigma_1 < \sigma_2 < \sigma_3$$

# Cryopumping Basics . . . Sticking Coefficients

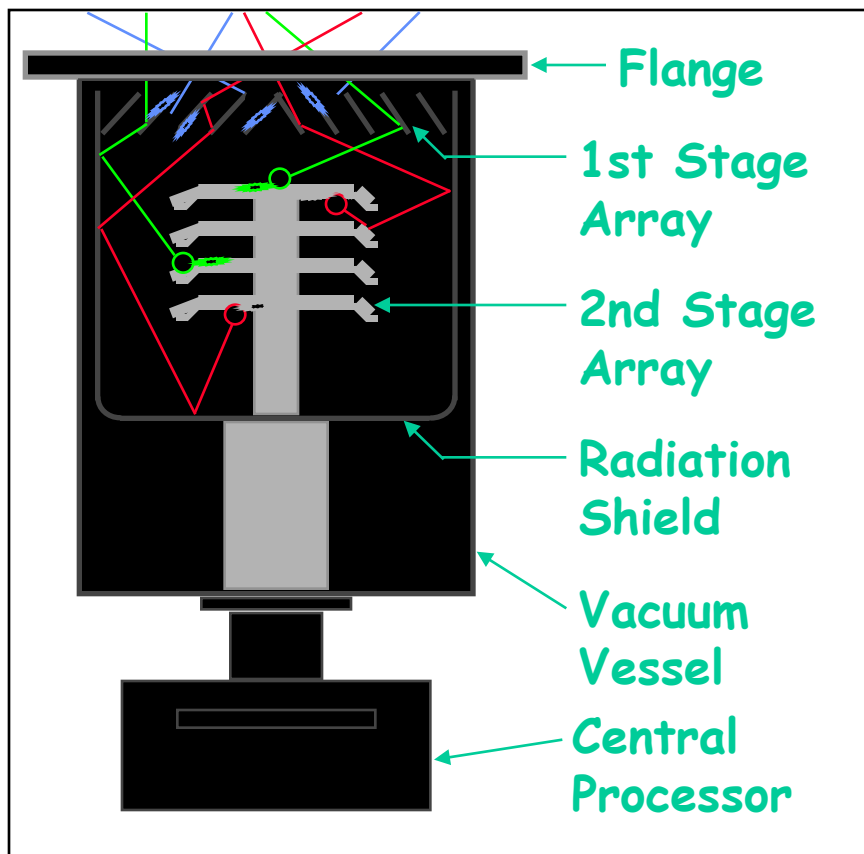


CryoSurface Temperature (K)	Gas and Temperature									
	N <sub>2</sub>		CO		O <sub>2</sub>		Ar		CO <sub>2</sub>	
	77 K	300 K	77 K	300 K	77 K	300 K	77 K	300 K	77 K	300 K
10	1.0	0.65	1.0	0.90			1.0	0.68	1.0	0.75
12.5	0.99	0.63	1.0	0.85			1.0	0.68	0.98	0.70
15	0.96	0.62	1.0	0.85			0.90	0.67	0.96	0.67
17.5	0.90	0.61	1.0	0.85	1.0	0.86	0.81	0.66	0.92	0.65
20	0.84	0.60	1.0	0.85			0.80	0.66	0.90	0.63
22.5	0.80	0.60	1.0	0.85			0.79	0.66	0.87	0.63
25	0.79	0.60	1.0	0.85			0.79	0.66	0.85	0.63
77									0.85	0.63

Ref. "Cryopumping", Dawson and Haygood, *Cryogenics* 5 (2), 57, (1965)



# Cryopump



Capture Type Pump

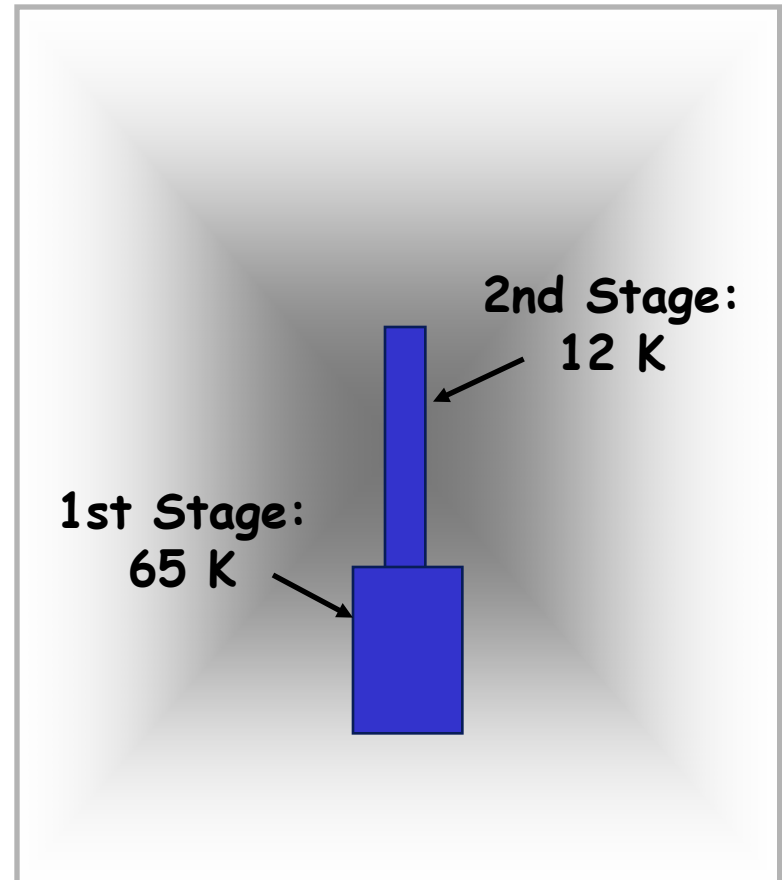
## Characteristics:

- No fluids, lubricants, or (in-vacuum) moving parts
- High crossover capability minimizes back-streaming
- High water pumping speed
- Tailorable pumping speeds
- Operate in all orientations
- Continuous backing not required

# Cryopump Components . . . *The Cold-Head*



- A cryopump is built around the **cold-head**.
  - Creates the cold temperatures needed to condense and adsorb gases
  - Two stages, each at a different temperature
- Achieves these temperatures by the expansion of helium.

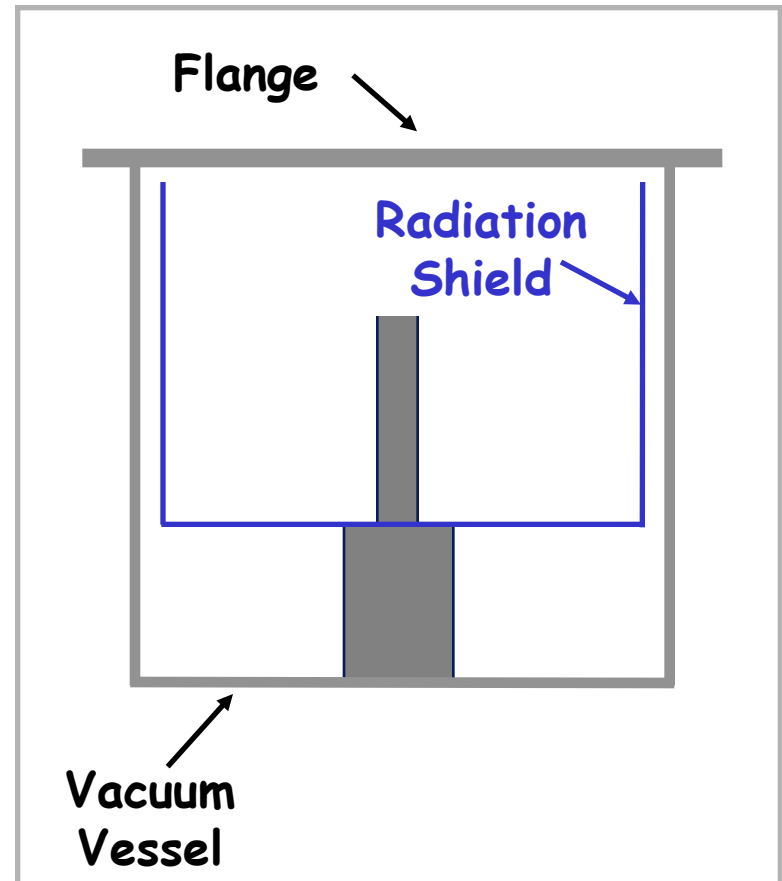


# Cryopump Components . . .

*shield, vacuum vessel, and flange*



- A **radiation shield** is attached to the 1st stage of the cold-head.
  - Copper for conductivity
  - Nickel plating for protection
- The **vacuum vessel** isolates the cryopump.
- The inlet **flange** attaches to the chamber.

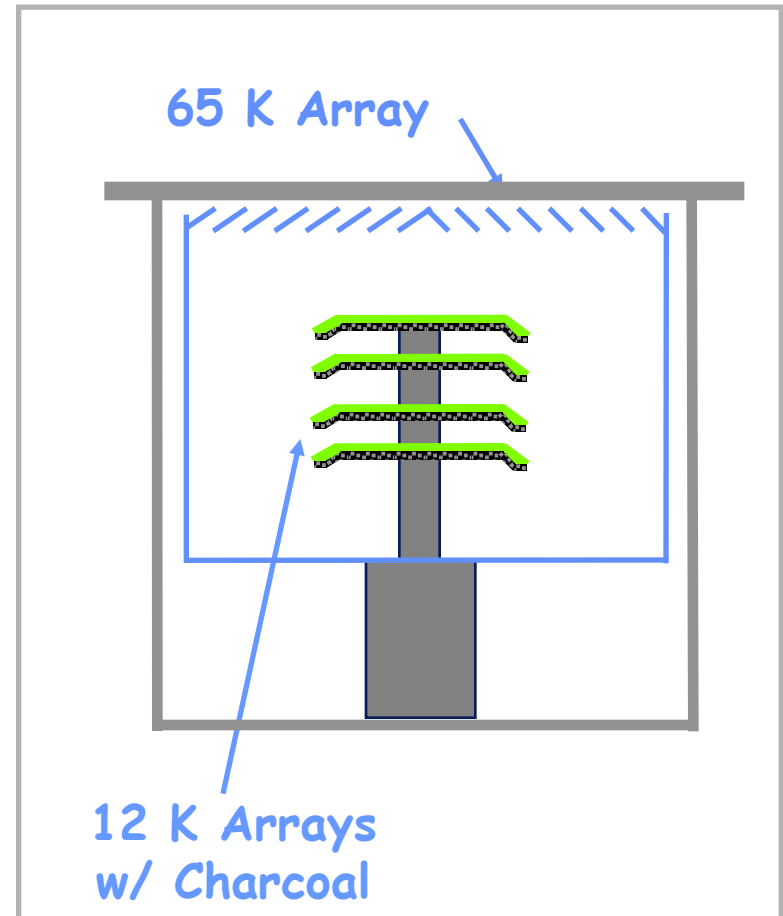


# Cryopump Components . . .

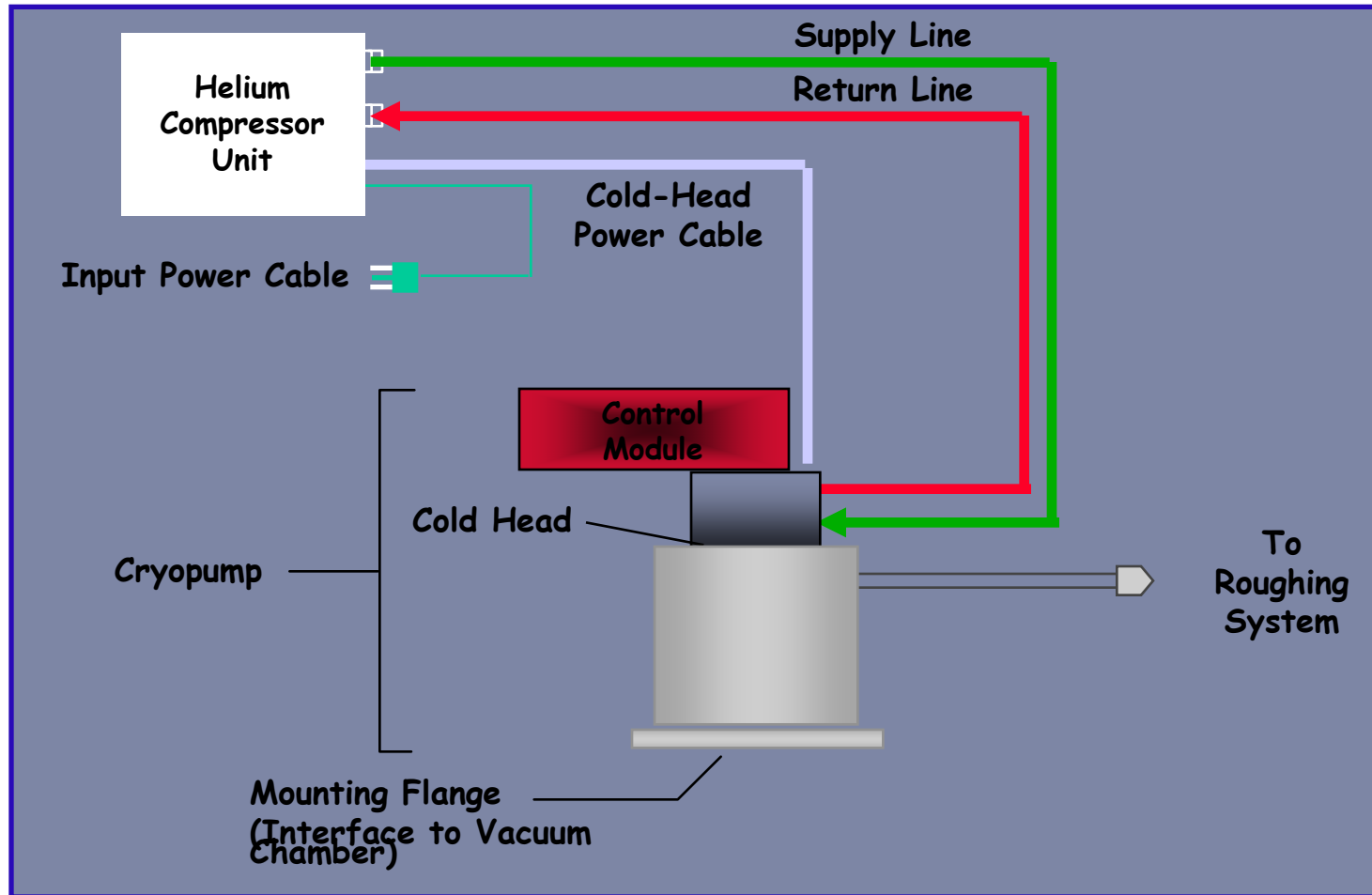
## *1<sup>st</sup> and 2<sup>nd</sup> Stage Arrays*



- The **1st stage** (65 K) array is attached to the radiation shield.
  - Condenses water vapor
- A series of arrays with charcoal are attached to the **2nd stage** (12 K) of the cold-head.
  - Condenses O<sub>2</sub>, N<sub>2</sub>, Ar
  - Adsorbs H<sub>2</sub>, He, Ne



# Cryopump System Overview

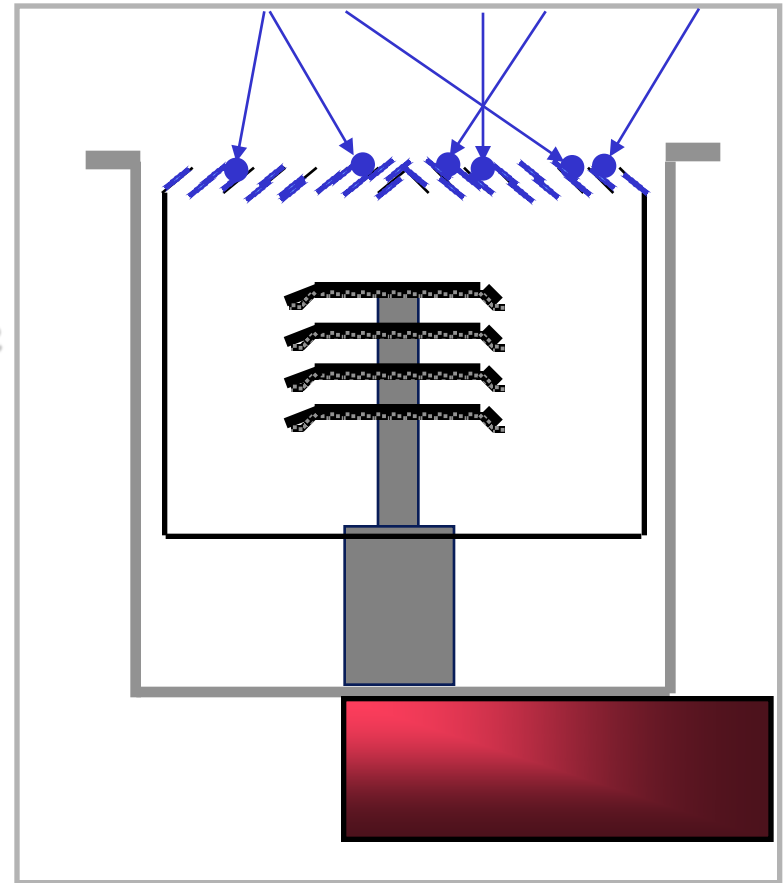




# Cryopump Operation - *Cryocondensation*



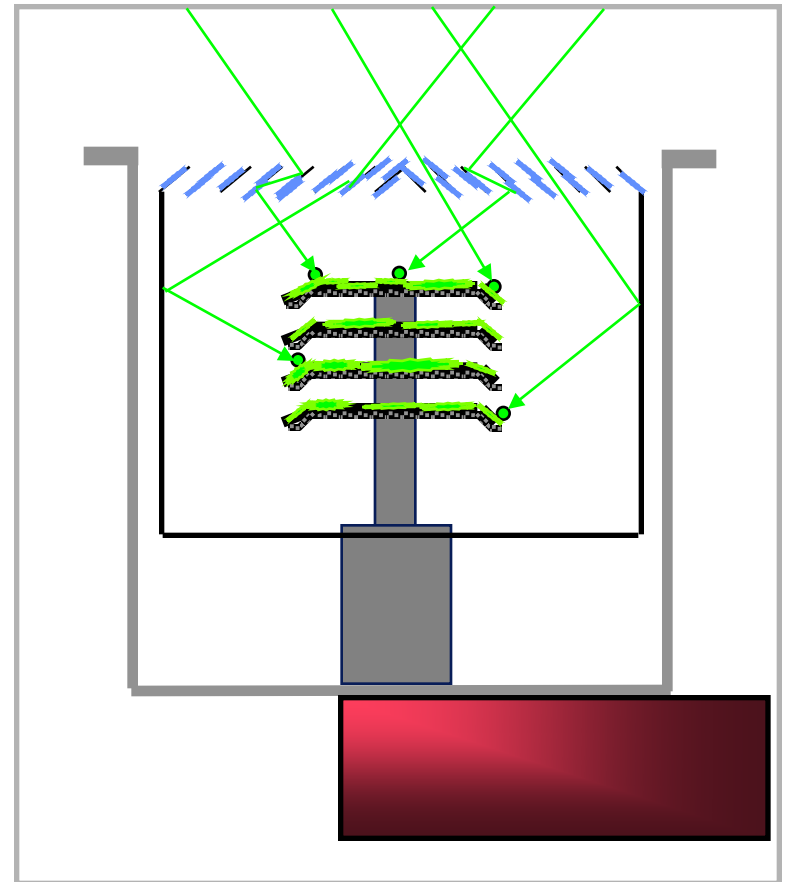
- Water molecules collide with the cooled surfaces of the 65 K first stage array.
- Condensation layers form as more of these molecules collect.



# Cryopump Operation - *Cryocondensation*



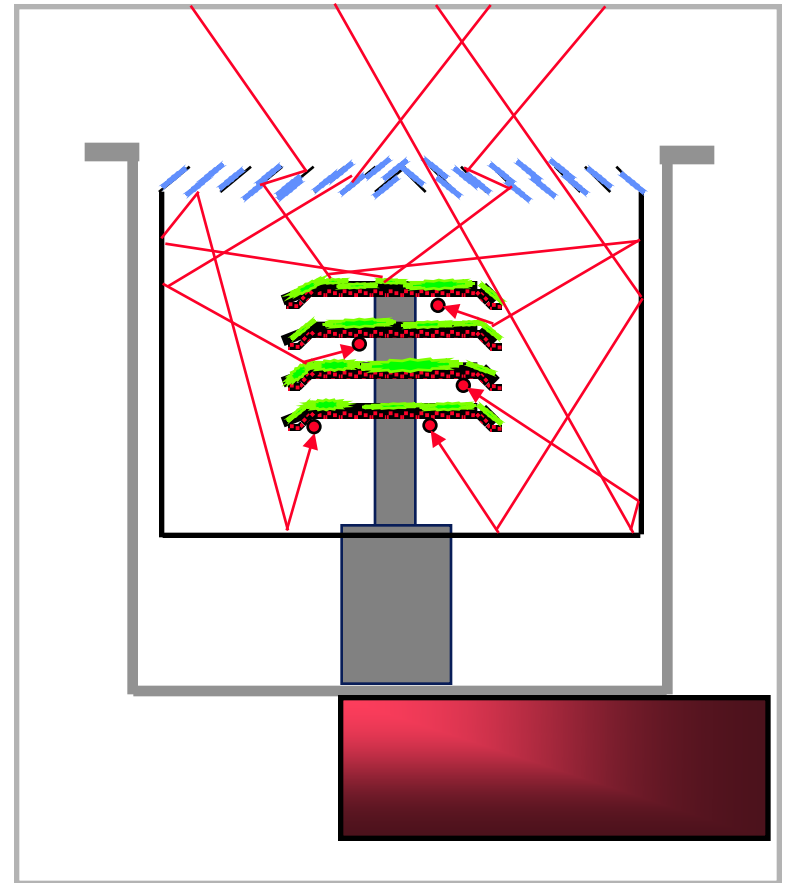
- Other molecules such as oxygen, nitrogen, and argon pass between the first stage arrays.
- By colliding with the 12 K second stage arrays, these molecules also form condensation layers.



# Cryopump Operation - Cryoadsorption



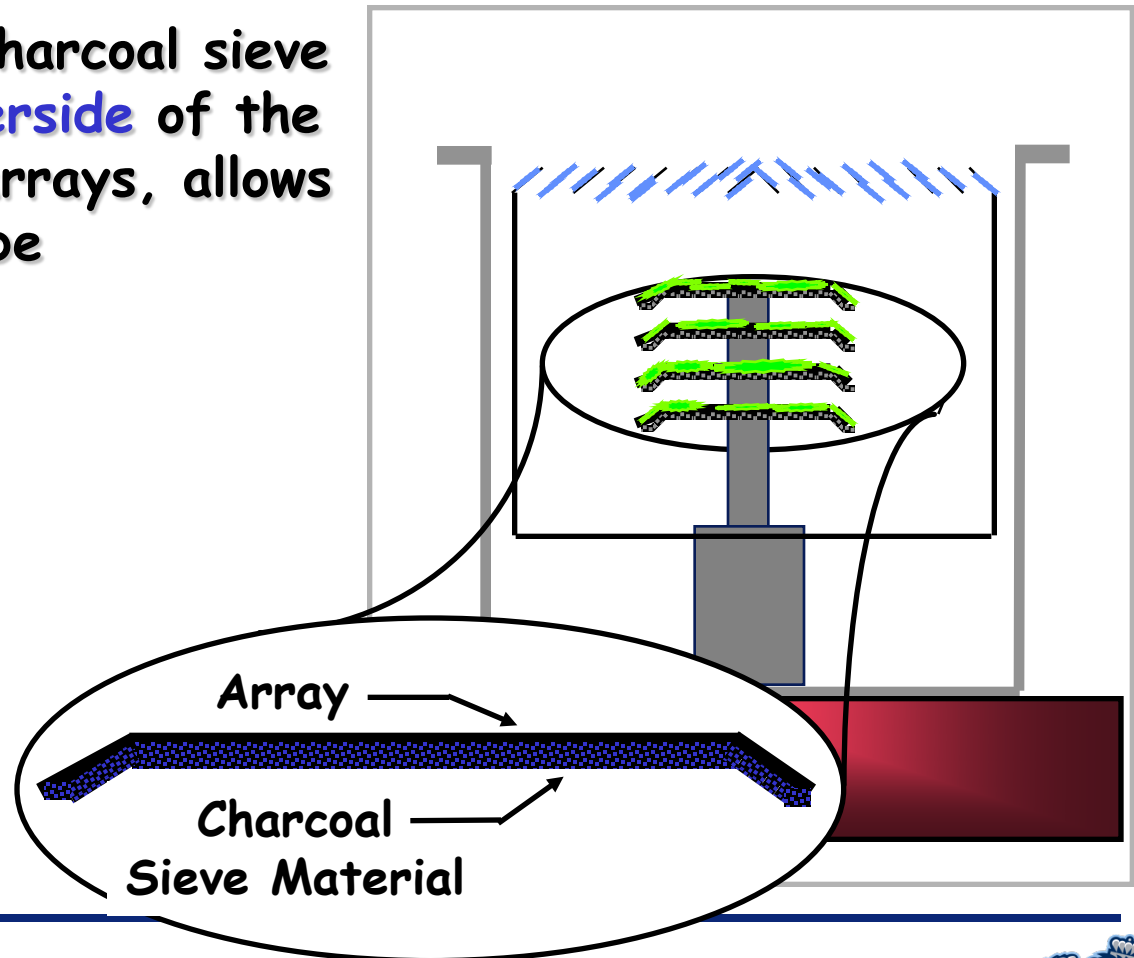
- The noncondensable  $H_2$ , He, and Ne molecules pass between the first stage arrays.
- Collide with walls and second stage arrays.
- Become adsorbed upon contacting the charcoal surfaces.





# Cryopump Operation - Cryoadsorption

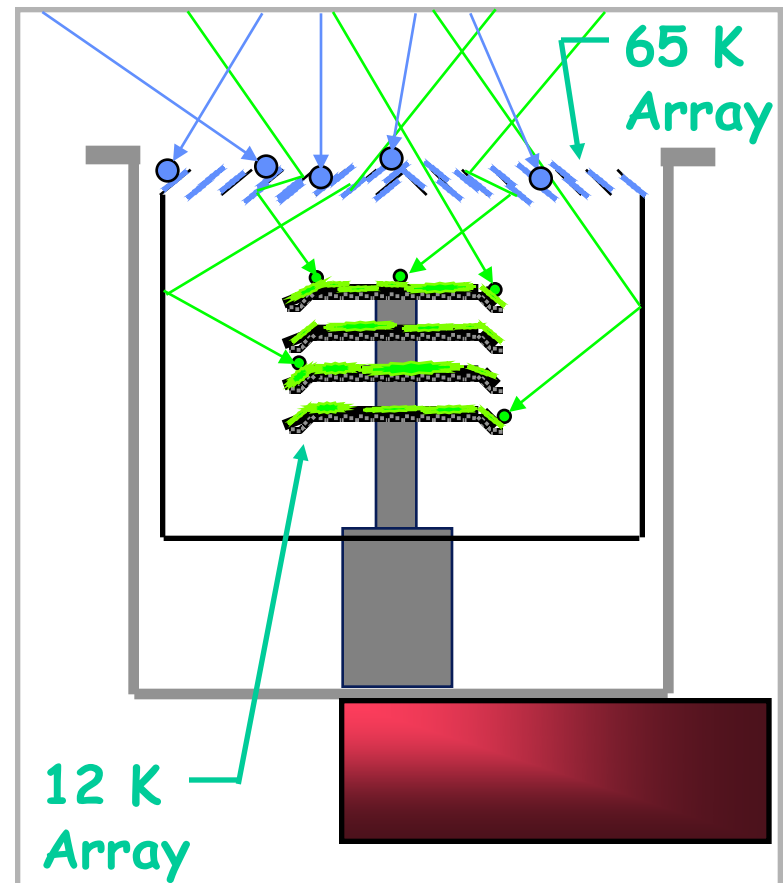
- Affixing activated charcoal sieve material to the **underside** of the 12 K second stage arrays, allows  $H_2$ , He, and Ne to be cryoadsorbed.



# Cryopump Operation - Argon Hang-Up



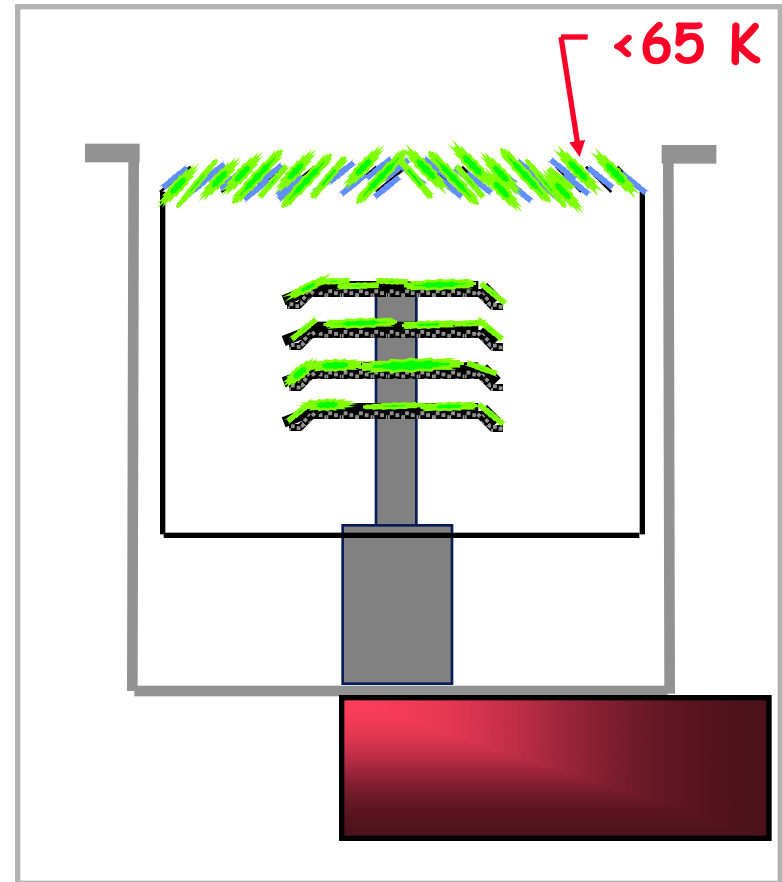
During normal operation, water vapor is condensed on the 65 K first stage array while oxygen, nitrogen, and argon are condensed on the 12 K second stage array.





# Cryopump Operation - Argon Hang-Up

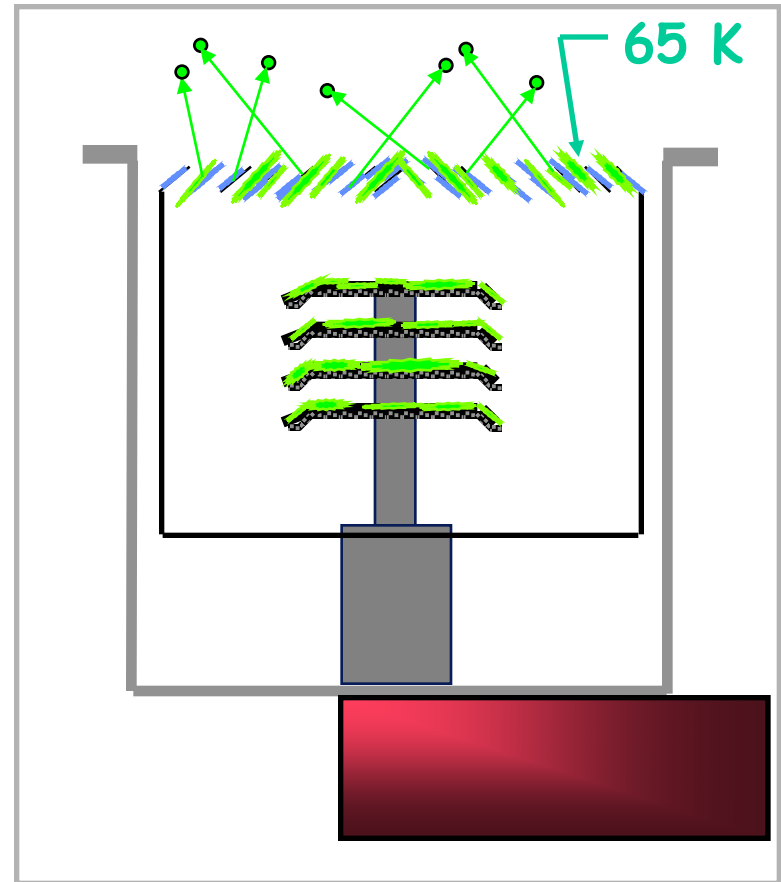
- Argon Hang-Up can occur if the first stage gets too cold.
- Results in argon being condensed (pumped) on the first stage.
- Where it stays until lower partial pressures are reached.



# Cryopump Operation - Argon Hang-Up



- When the equilibrium pressure is reached.
  - Argon liberates
  - Pumpdown slows
  - Causes "False Full" condition



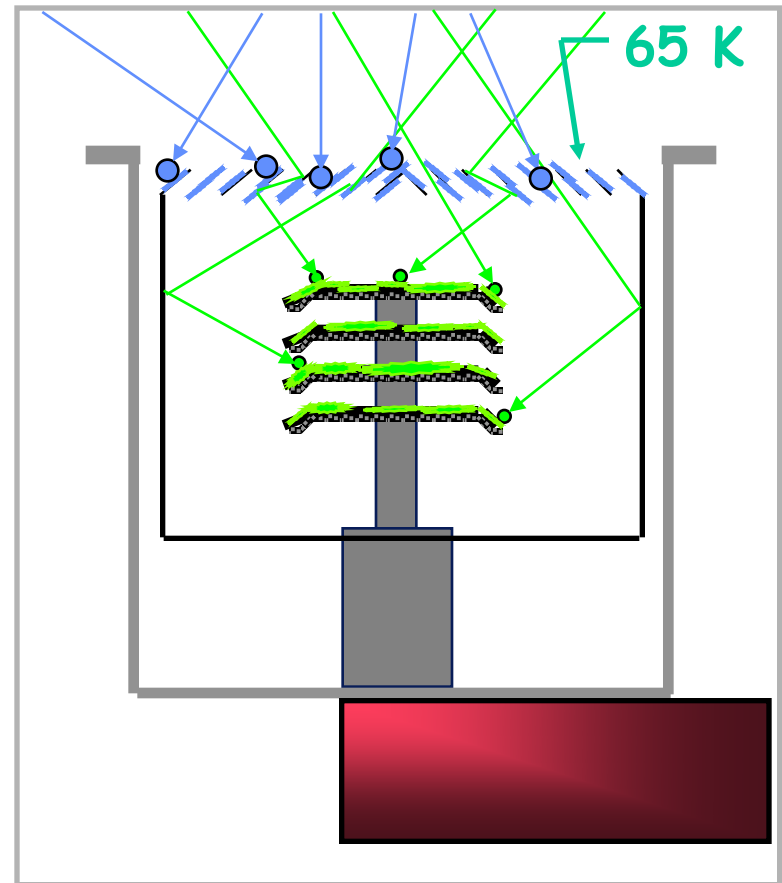
## EQUILIBRIUM VAPOR PRESSURE

	$10^{-10}$	$10^{-7}$	$10^{-4}$	$10^{-3}$
Water	130K	153K	185K	198.5K
Argon	23.7K	28.6K	35.9K	39.2K



# Cryopump Operation - Argon Hang-Up

- Argon liberates until it is repumped onto the second stage where it should have been pumped.

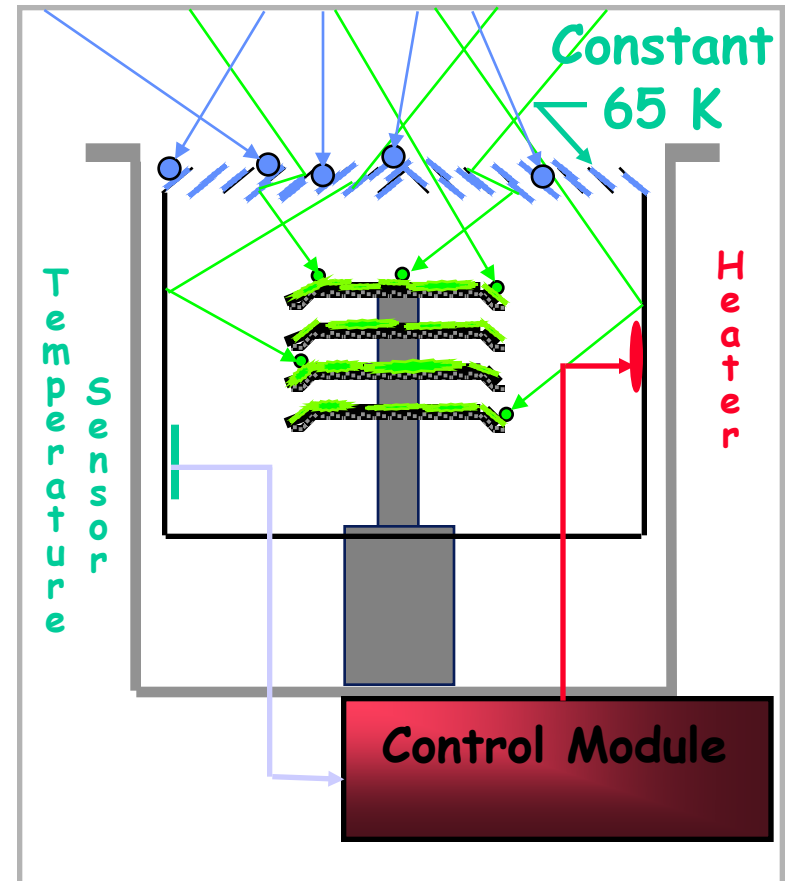




# Cryopump Operation - Argon Hang-Up



- Argon Hang-Up can be avoided with modern controllers interfaced to the first stage sensor and heater.
  - Monitors and controls temperature
  - Prevents a "Too Cold" condition



# Cryopump Example Parameters



Parameter	Value
Pumping speed (l/s) water	9000–10500
Air	3000–3250
Hydrogen	4500–5200
Argon	2500–2700
Helium	1500–2300
Maximum throughput (Pa·m <sup>3</sup> /s) argon	1.0–2.5
Hydrogen	1.2
Pumping capacity (Pa·m <sup>3</sup> ) argon	$1.5 \times 10^5$ – $3 \times 10^5$
Hydrogen	1500–5000
Helium	10–100
Ultimate pressure (N <sub>2</sub> equivalent) (Pa)	$10^{-9}$ – $10^{-10}$
Cool-down time (h)	1.5–2.5
Crossover (Pa·m <sup>3</sup> )	35–50
Weight (kg)	30–50

*Listed performance data (averaging from the catalogue of different manufacturers) for a typical 3000 l/s class two-stage cryopump*



# Cryopump Operation . . . *Crossover*

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During chamber evacuation, when should the high-vacuum valve be opened?

For cryopumps, the maximum crossover capability is specified as the impulsive mass input that causes the second stage to rise no higher than 20 K.

# Cryopump Operation . . . *Crossover*

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## Example: Crossover Pressure Calculation

Crossover value for a CTI On-Board 8 = 150 Torr-liters

**Crossover formula:**  $\frac{\text{Crossover value}}{\text{Chamber volume}} = P \text{ in Torr}$

$$\frac{150 \text{ Torr-liters}}{300 \text{ liters}} = .5 \text{ Torr or } 500 \text{ milliTorr}$$

Understanding crossover can produce faster  
pumpdown times and cleaner vacuum too.





# Cryopump Operation . . . *Regeneration*

The objective of regenerating a cryopump is to remove the captured gases from the pump and restore its pumping capacity.

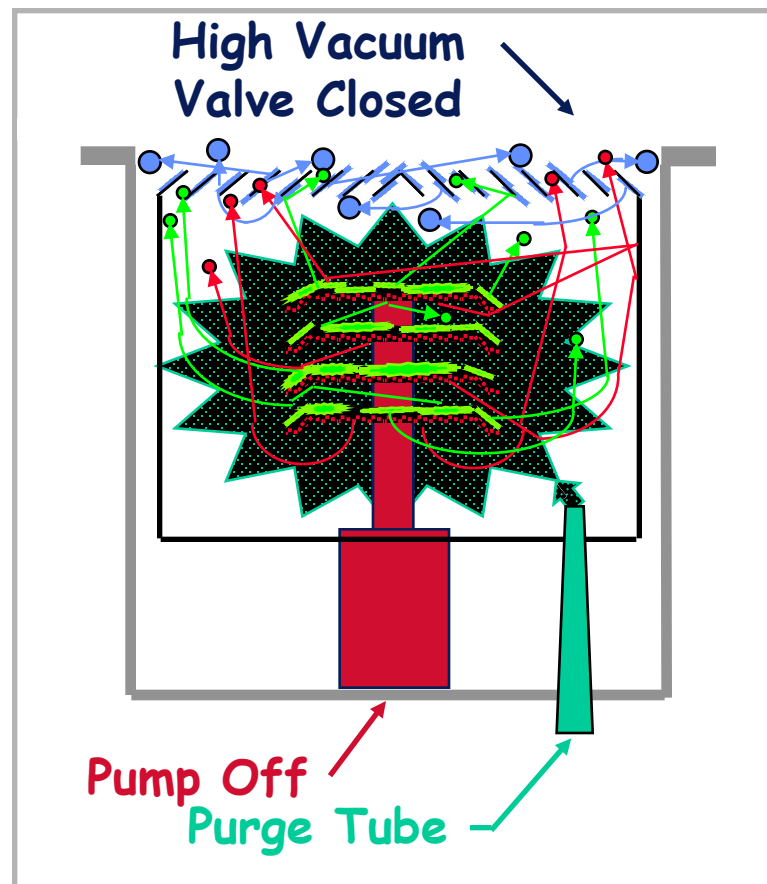
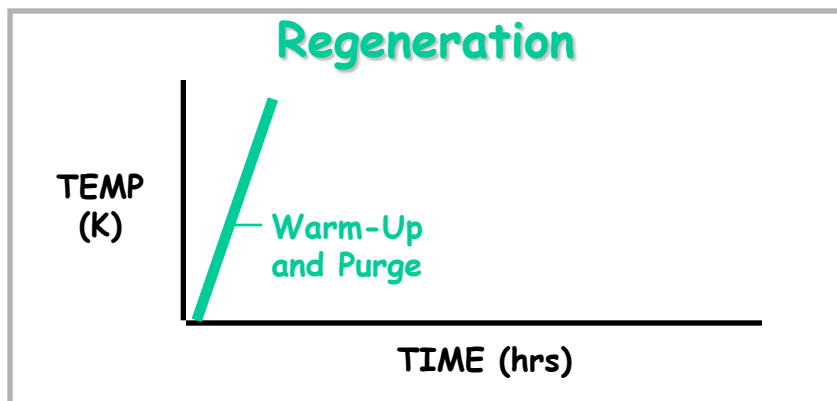
So . . . when should cryopumps be regenerated?

Whenever your system is down is a good opportunity to regenerate your cryopump without affecting your up-time.

# Cryopump Operation . . . Regeneration



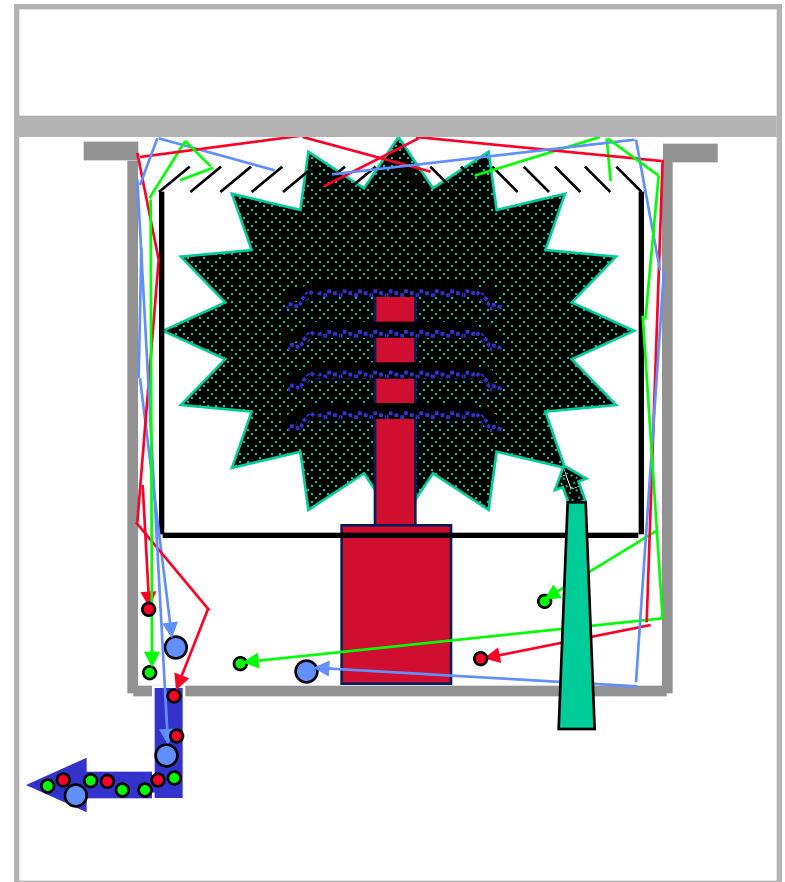
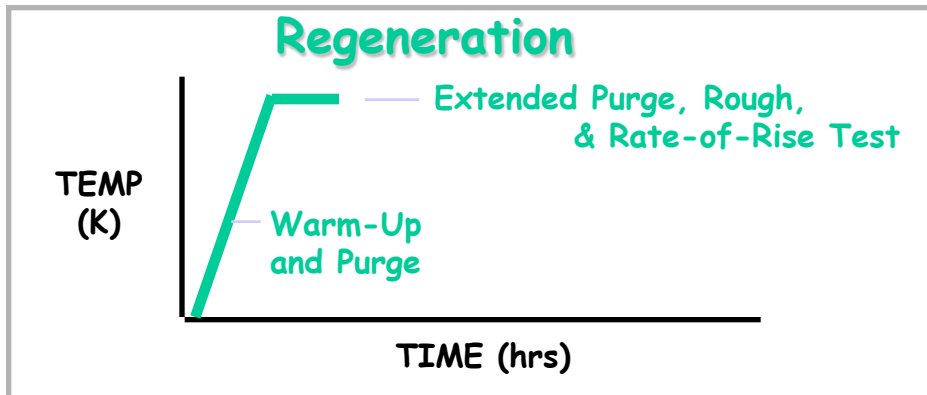
- Regeneration
  - Warm-Up and Purge



# Cryopump Operation . . . Regeneration



- **Regeneration**
  - Warm-Up and Purge
  - Extended Purge
  - Rough Out
  - Rate-of-Rise (ROR) Test

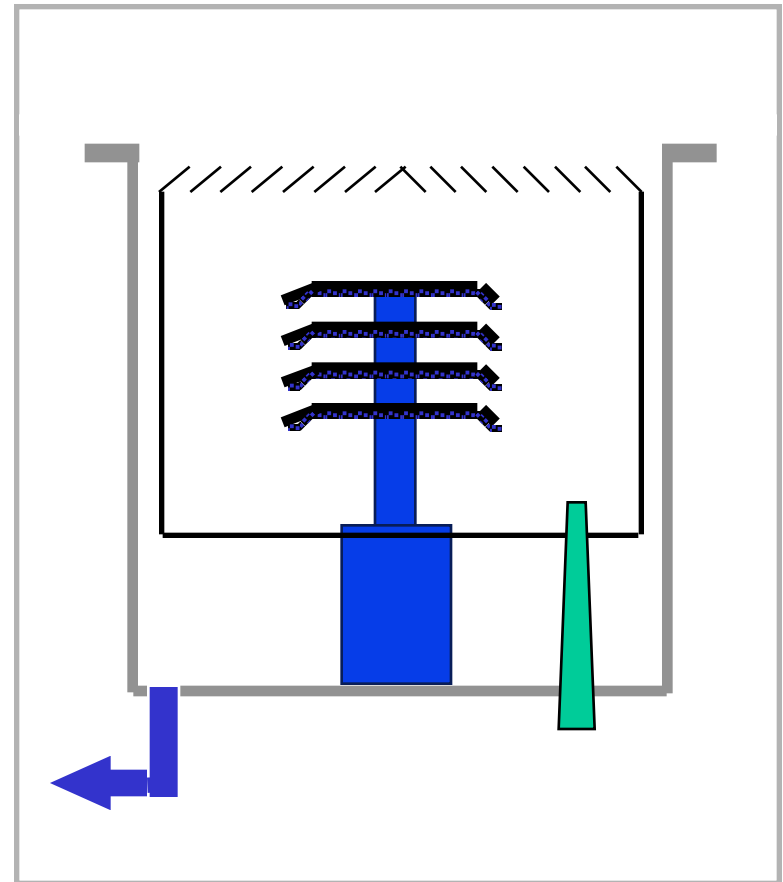
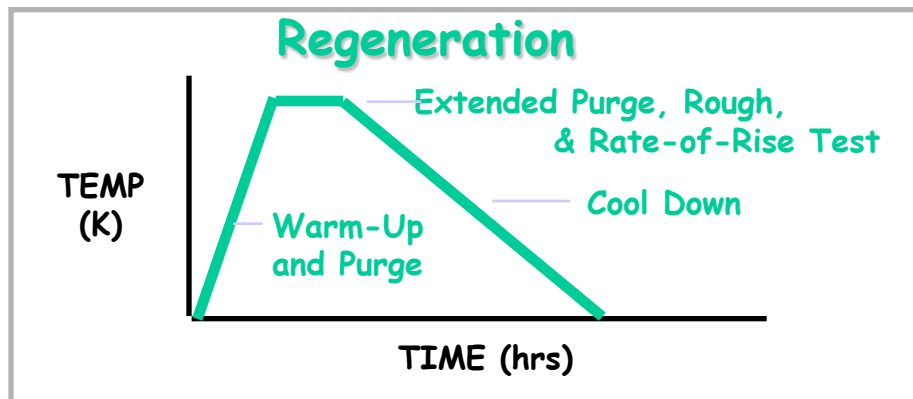




# Cryopump Operation . . . *Regeneration*

- **Regeneration**

- Warm-Up and Purge
- Extended Purge
- Rough Out
- Rate-of-Rise (ROR) Test
- **Cool Down**



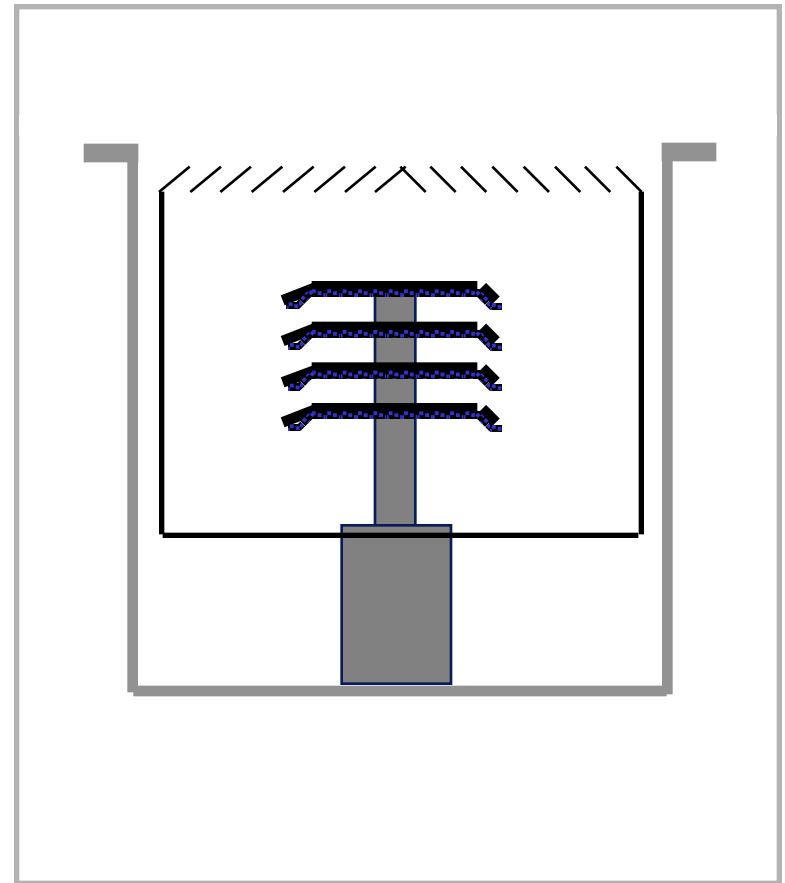
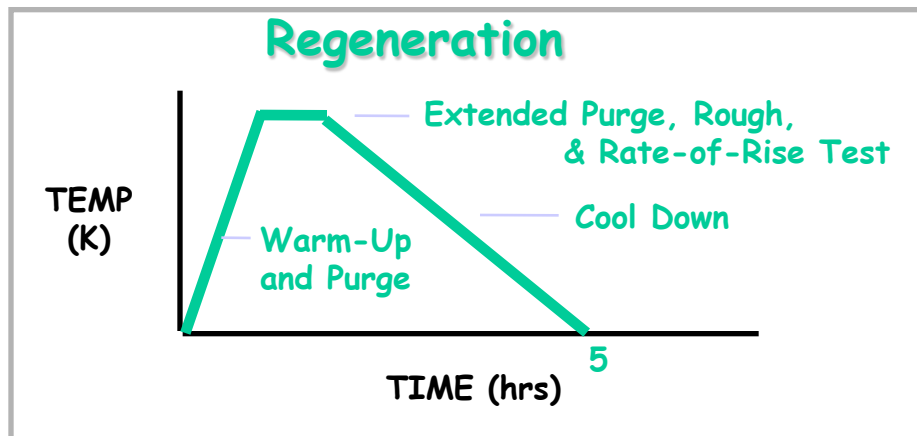


# Cryopump Operation . . . *Regeneration*

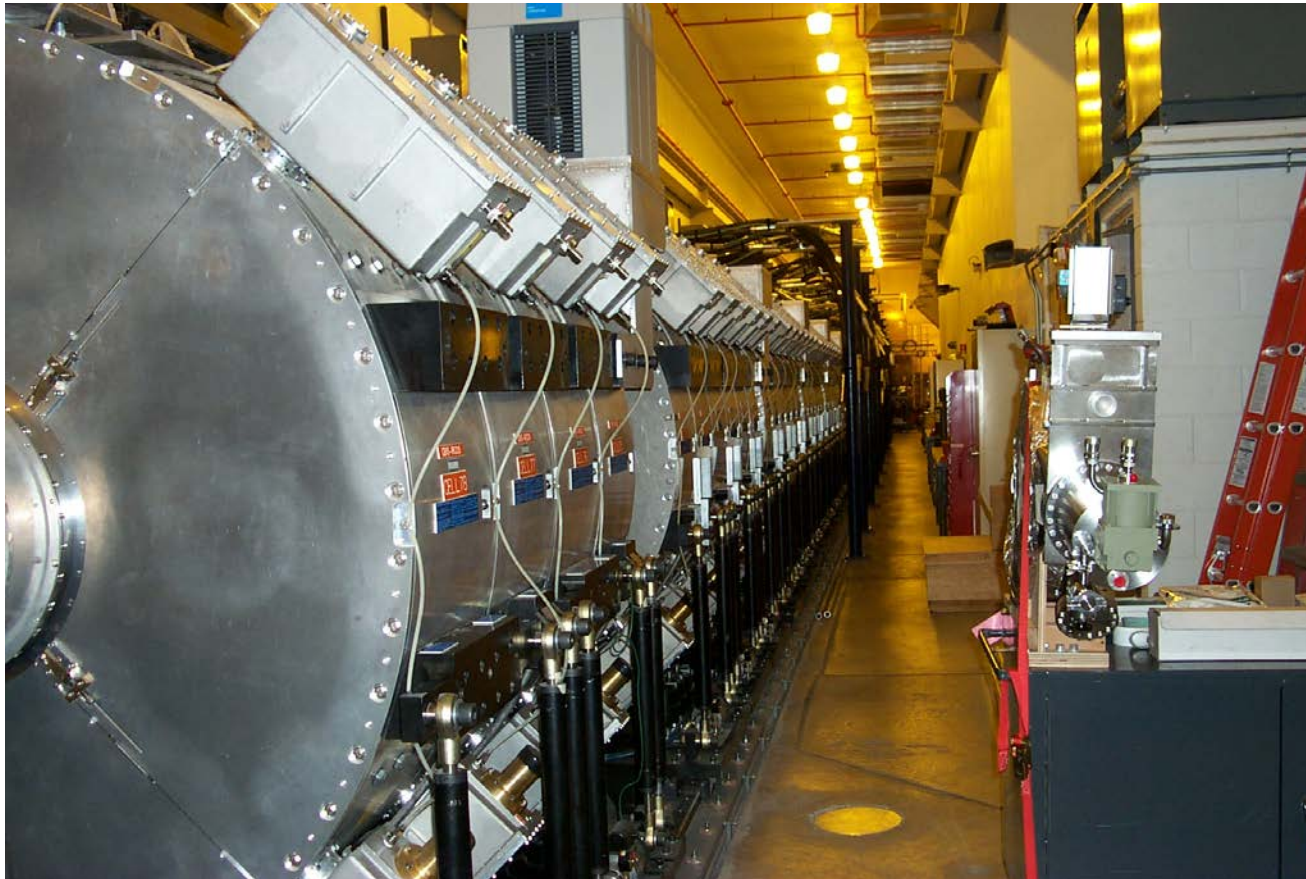


- Regeneration

Typically 5-6 hours cold-to-cold.



# Example of Cryo-pumped Accelerator - DARHT II (the Dual Axis Radiographic Hydro-Test)



# Example of Cryopumped Accelerator - APT RFQ



- **Cryogenic Pumping System for Cavity system, with H<sub>2</sub> Pumping Speed of 12,000 L/s**
- **This assembly was completed and successfully tested at LLNL Vacuum Lab. The whole system was then delivered and installed at the APT/LEDA facility.**

